

Treatment of Arsenic and Metals in Ground Water Using a Compost/ZVI PRB

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Columbia Nitrogen Site

tidal
marsh



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Columbia Nitrogen Site

- Site impacted by former phosphate fertilizer manufacturing operations
- Pyrite and elemental sulfur used in production of sulfuric acid
- Sulfuric acid stored in Pb-lined concrete vats
- Site extensively filled with operational materials and debris
- Ground water impacted by heavy metals (Pb, Cd), arsenic, and low pH





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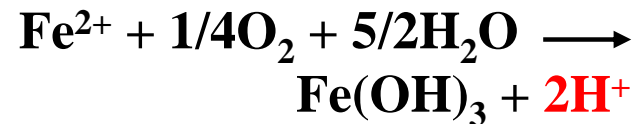
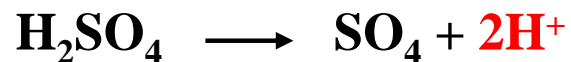
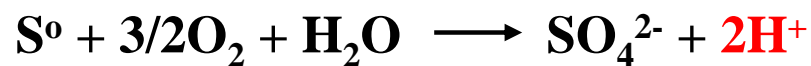
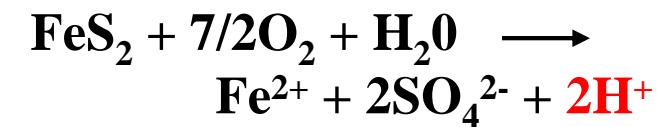


Current Scenario

sulfide/sulfur oxidation

iron oxidation

ground surface



tidal marsh

As
Pb
Cd
Fe²⁺
H⁺





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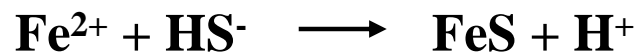
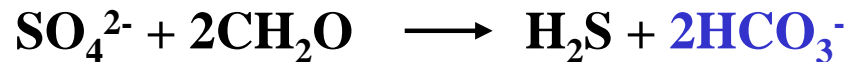
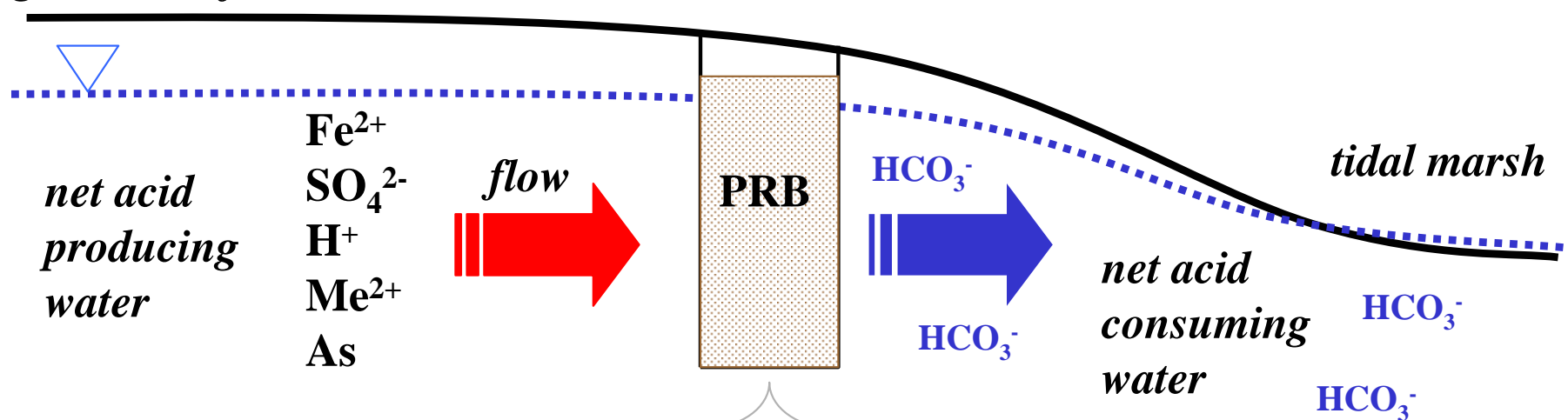
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PRB Treatment Objective

promote sulfate reduction

ground surface



PRB Treatment Objectives

- Raise pH of ground water
- Promote sulfate reduction to remove heavy metals, arsenic, and ferrous iron
- Prevent additional down-gradient ecosystem impacts by preventing acid production associated with ferrous iron oxidation
- Contribute to ecosystem restoration by converting ground water from acid producing to acid consuming system (i.e. generate an alkalinity plume)

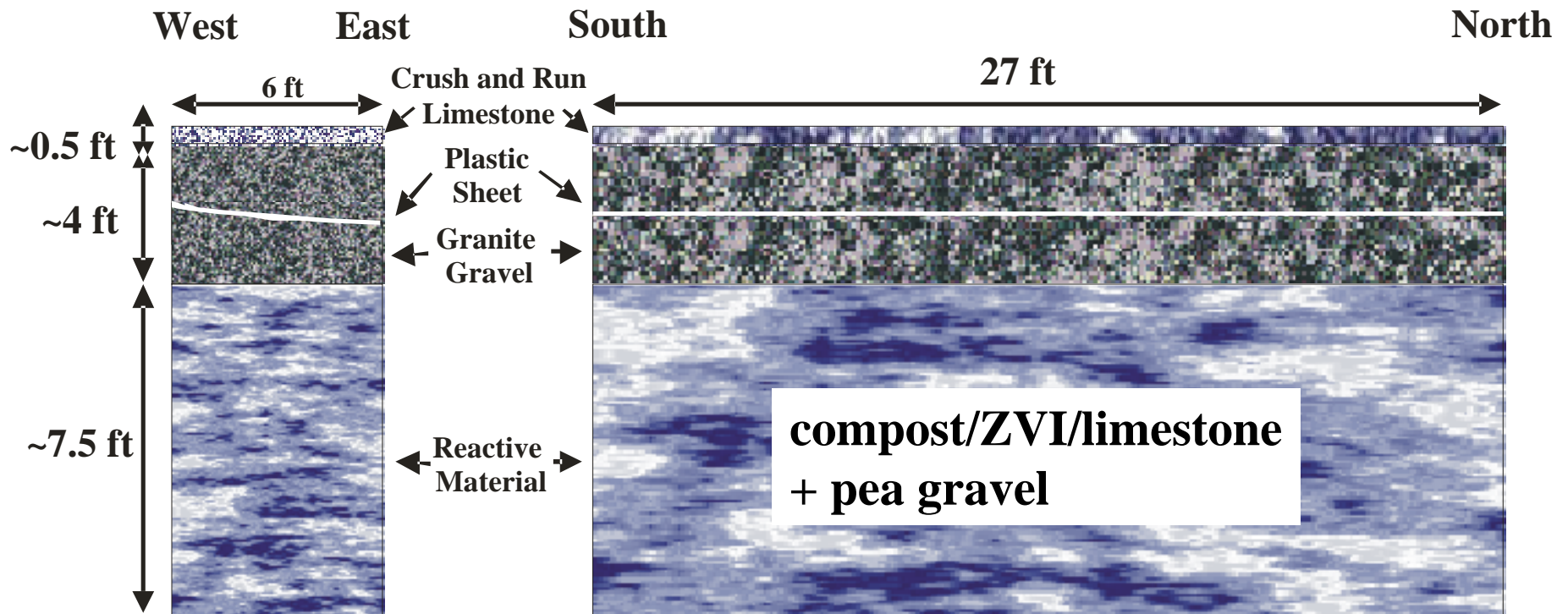


PRB Design Tests/Studies

- Laboratory column tests
 - Focused on optimizing reactivity and hydraulic conductivity
 - Mixture of 30% compost, 20% ZVI, 5% limestone and 45% pea gravel selected
- Permeameter tests
 - Three samples from final mixture tested for permeability to ensure permeability greater than that of aquifer
 - Result of permeability testing: $K = 6.7E-02$ cm/s



Vertical X-Section of Pilot PRB





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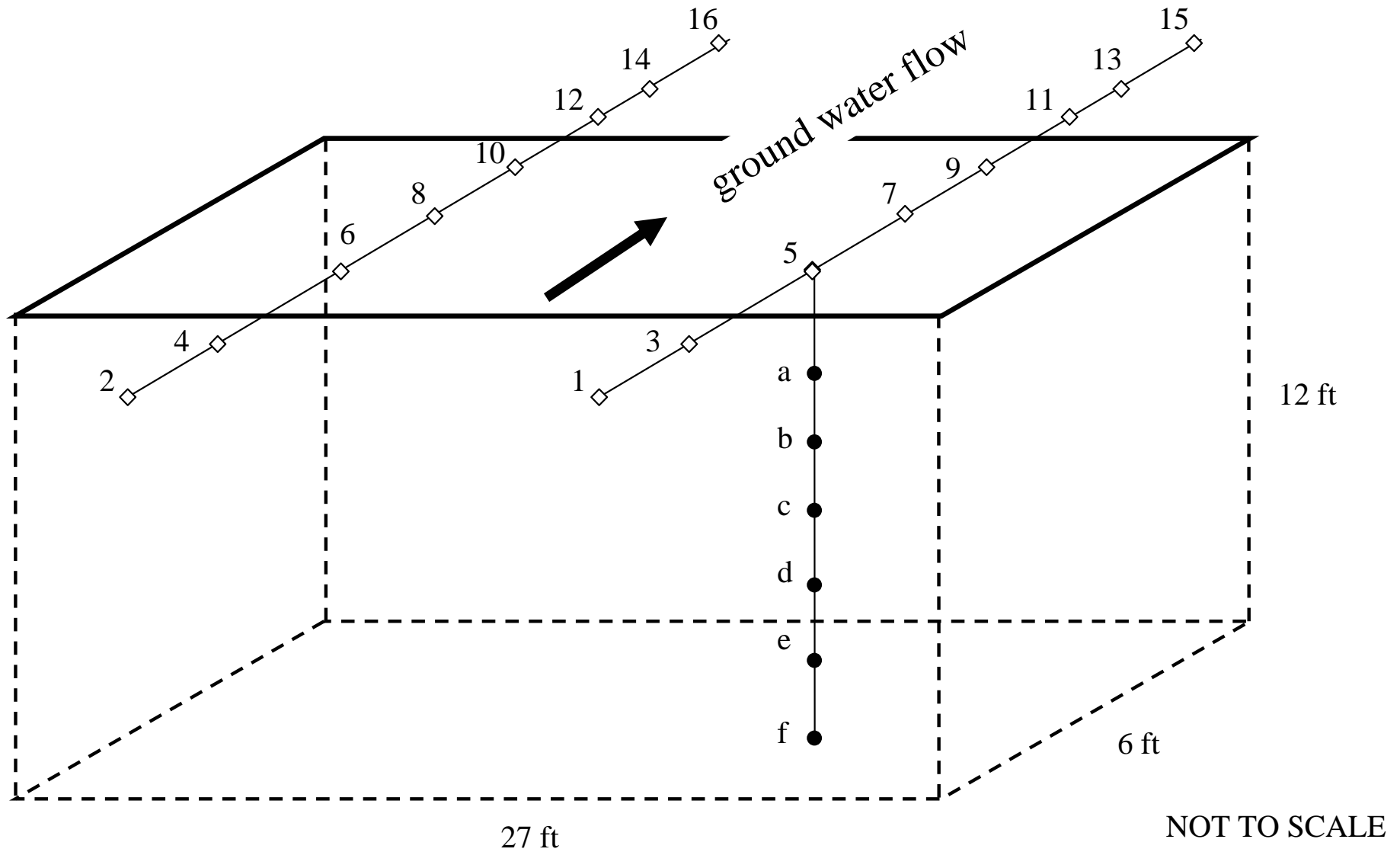


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Multi-Level Monitoring System Layout





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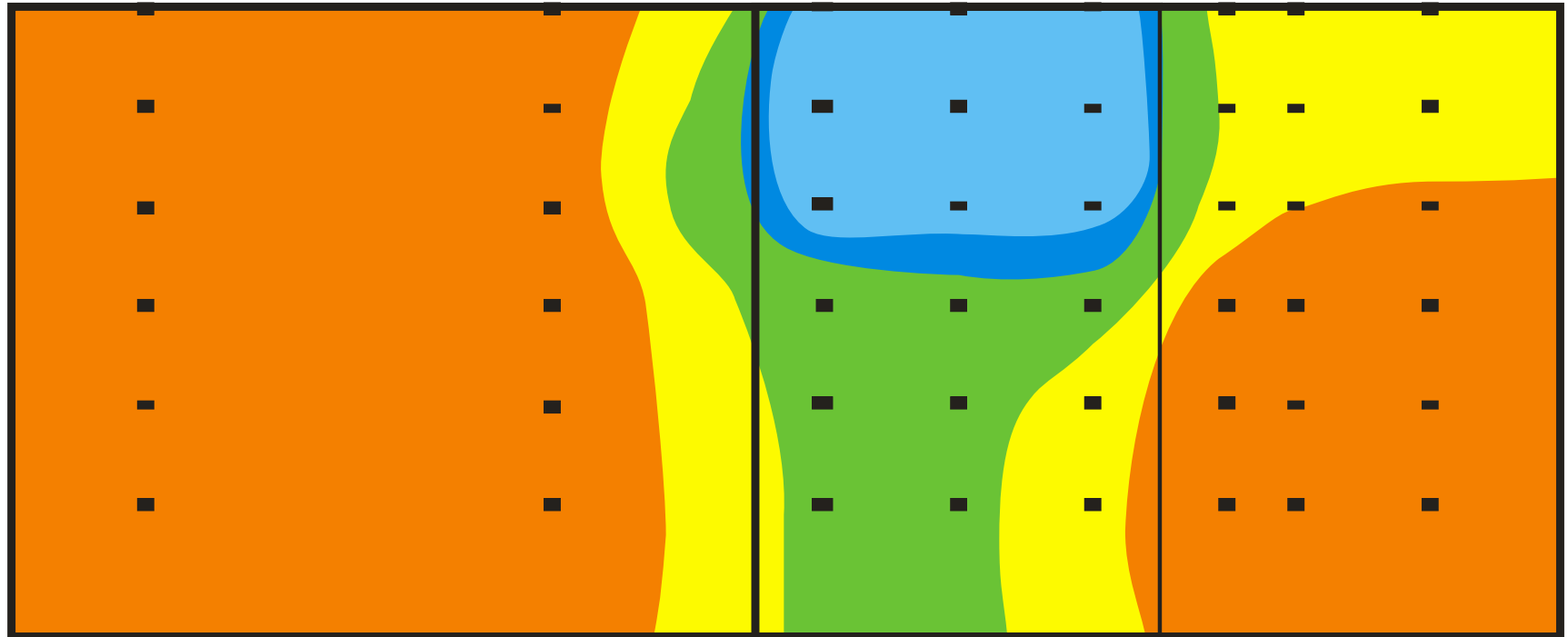
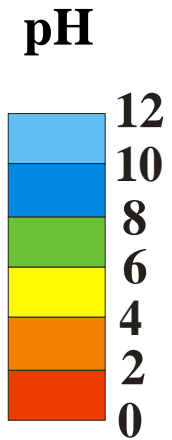
Performance Monitoring

- Chemical monitoring conducted 1, 3, 8, 13, and 18 months after installation
- Chemical parameters evaluated: Pb, As, Cd, Fe, pH, Eh, alkalinity, cations, anions, TOC/DOC, sulfide
- Microbial parameters: MPN sulfate reducers
- Hydraulic monitoring – slug tests



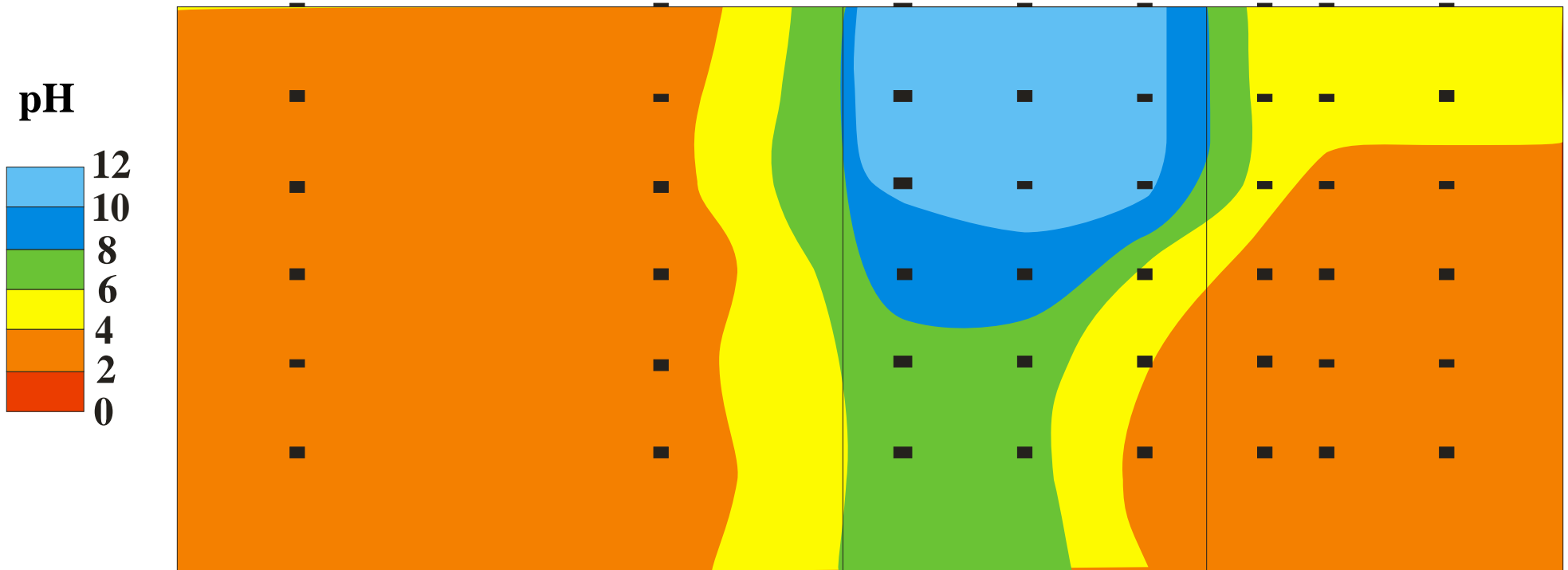
pH profile – 8 months

ground water flow →



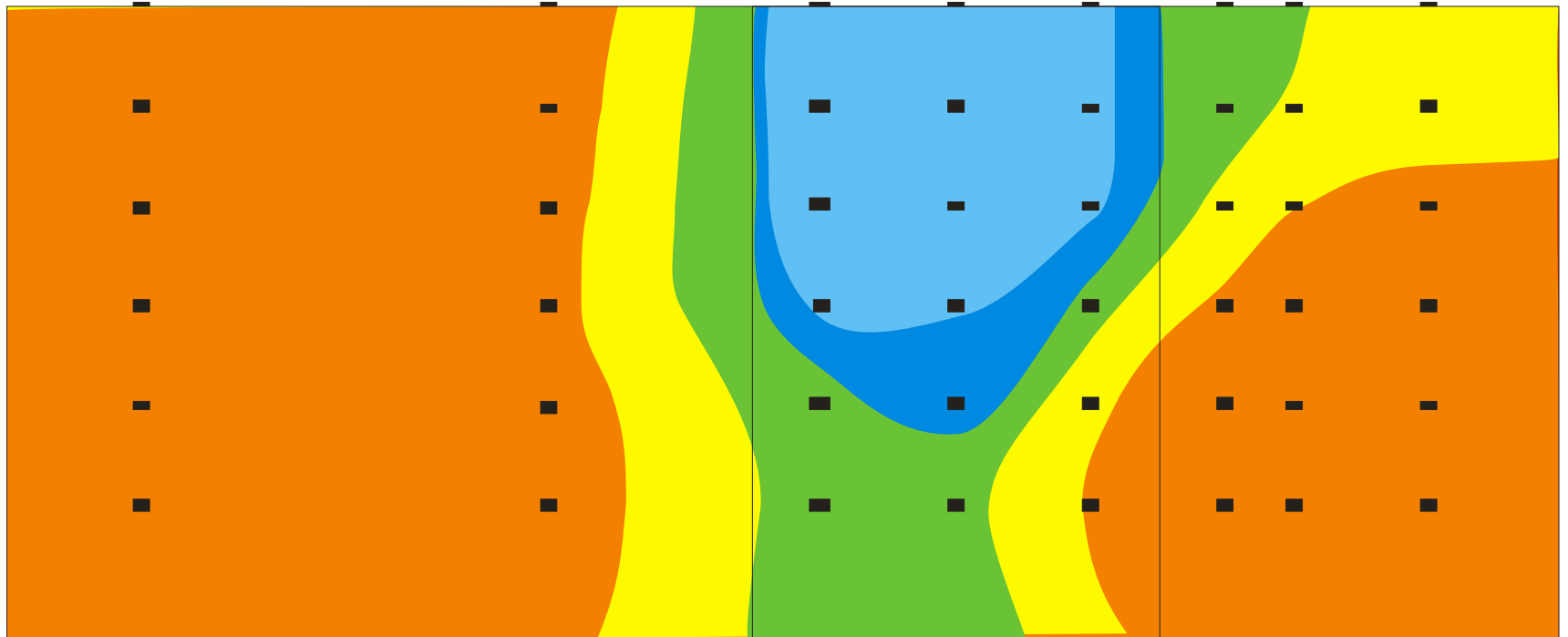
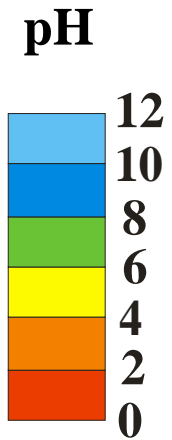
pH profile – 13 months

ground water flow →



pH profile – 18 months

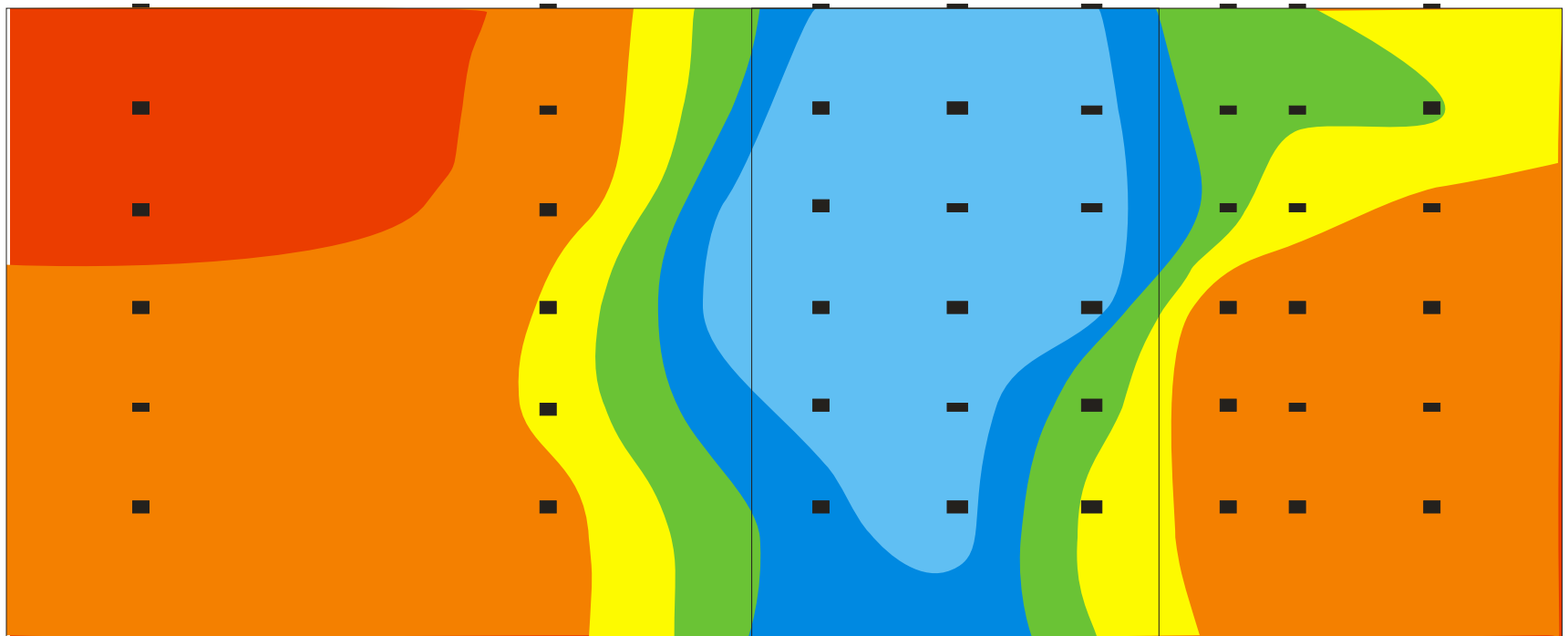
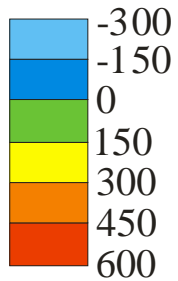
ground water flow →



Eh profile – 8 months

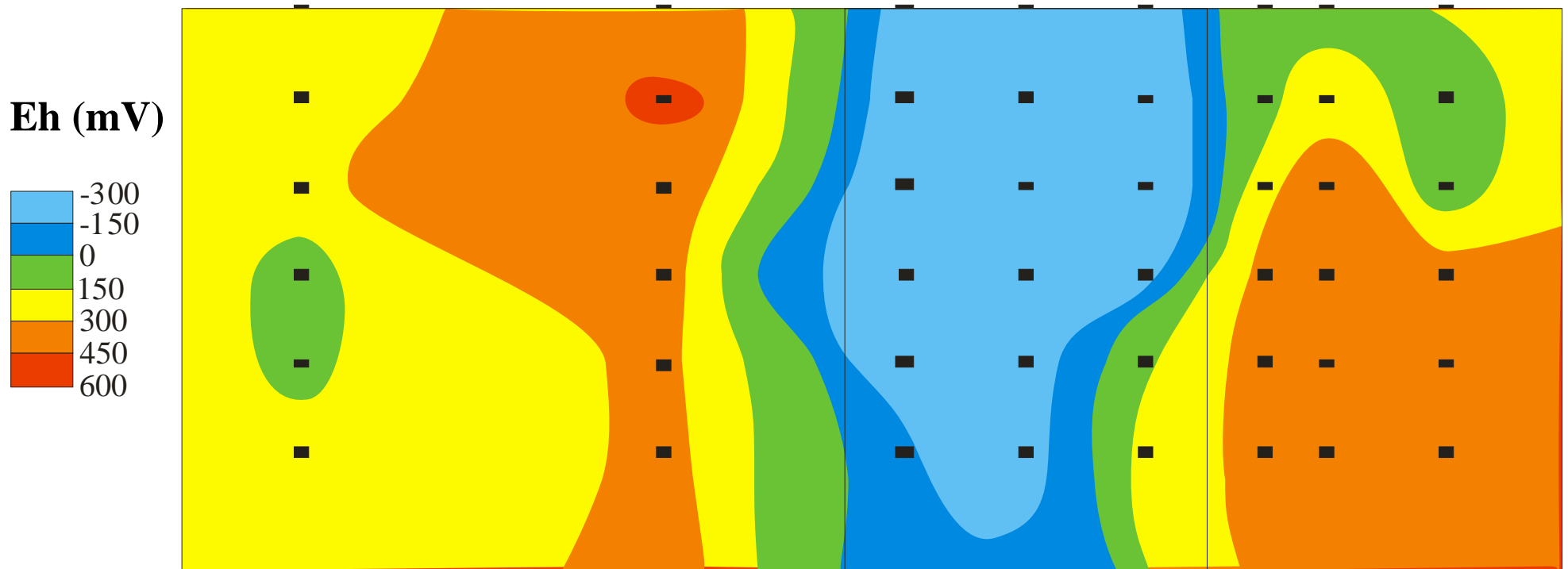
ground water flow 

Eh (mV)



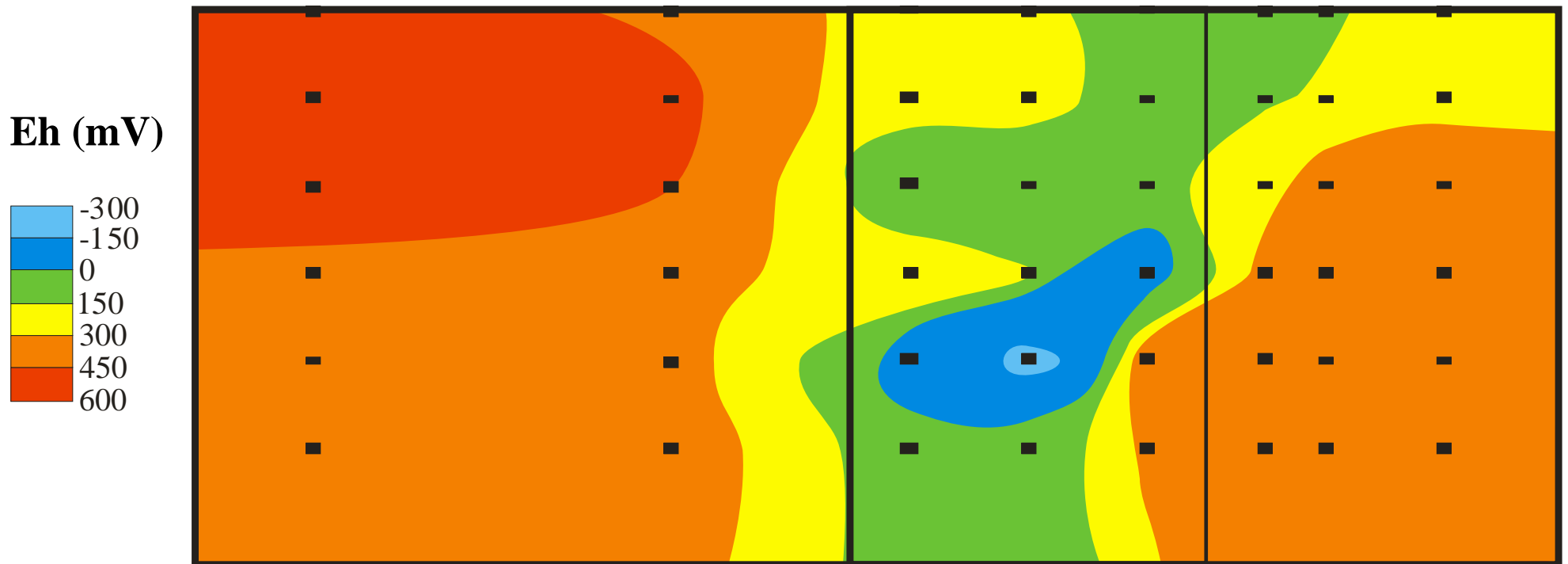
Eh profile – 13 months

ground water flow →



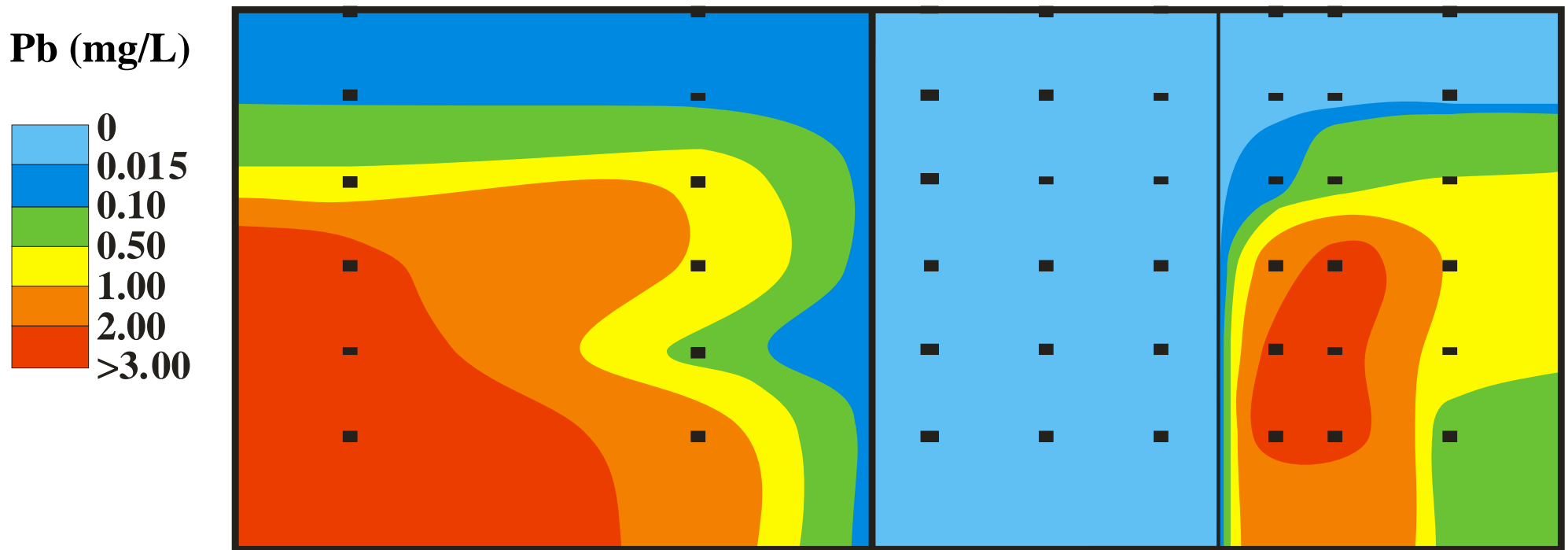
Eh profile – 18 months

ground water flow →

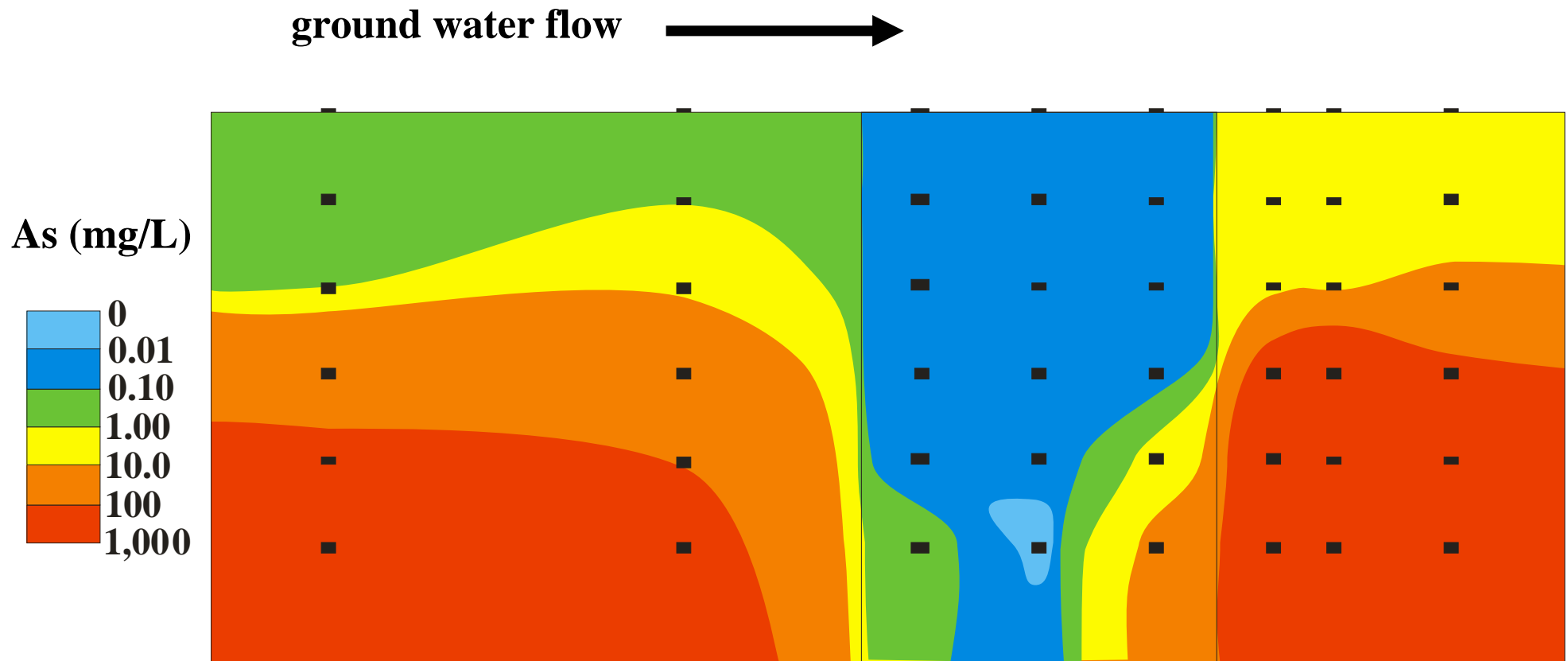


Pb concentration profile – 18 months

ground water flow 

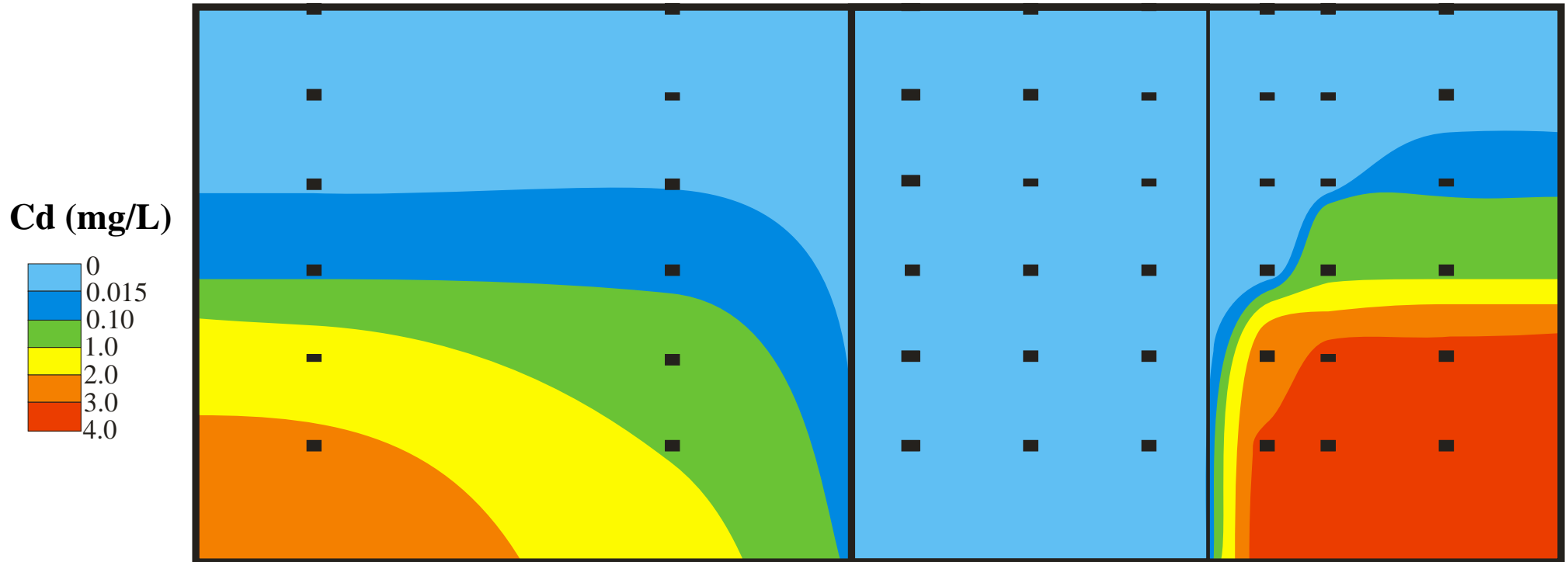


As concentration profile – 18 months



Cd Profile – 18 months

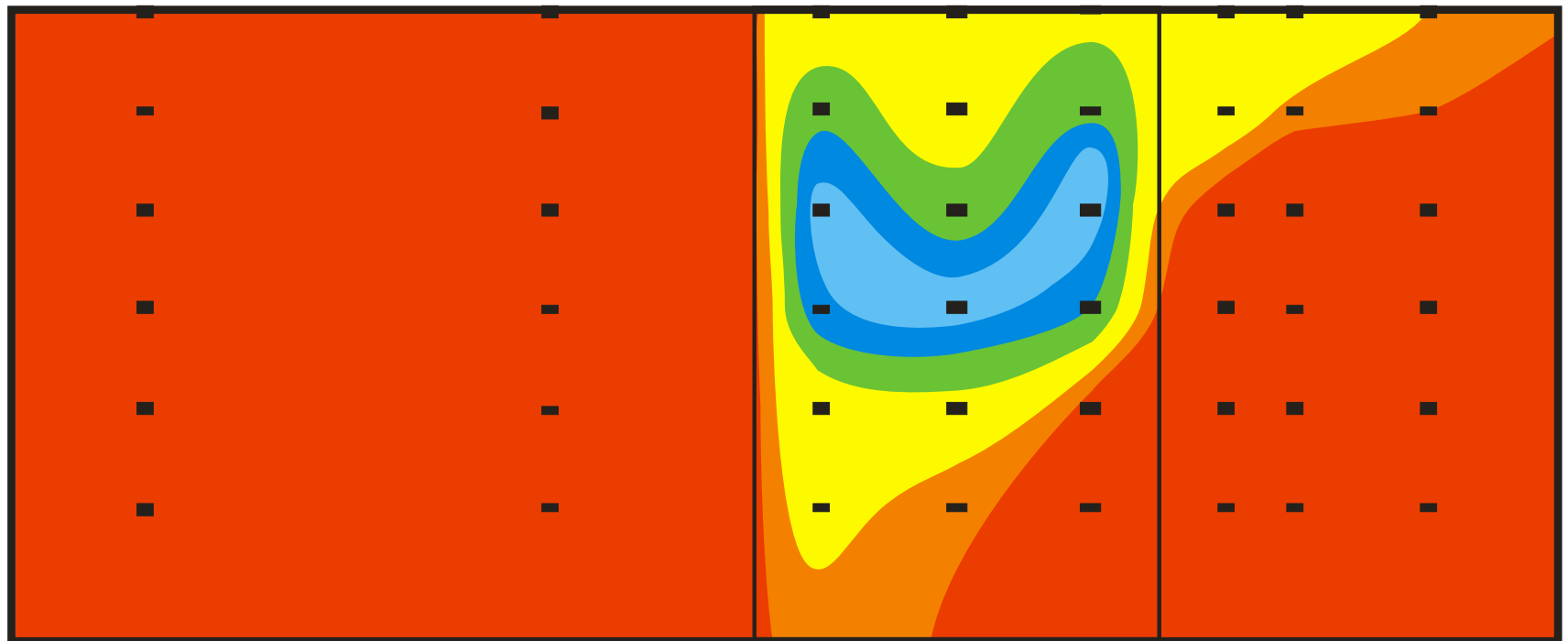
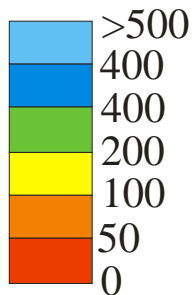
ground water flow →



Alkalinity profile – 18 months

ground water flow →

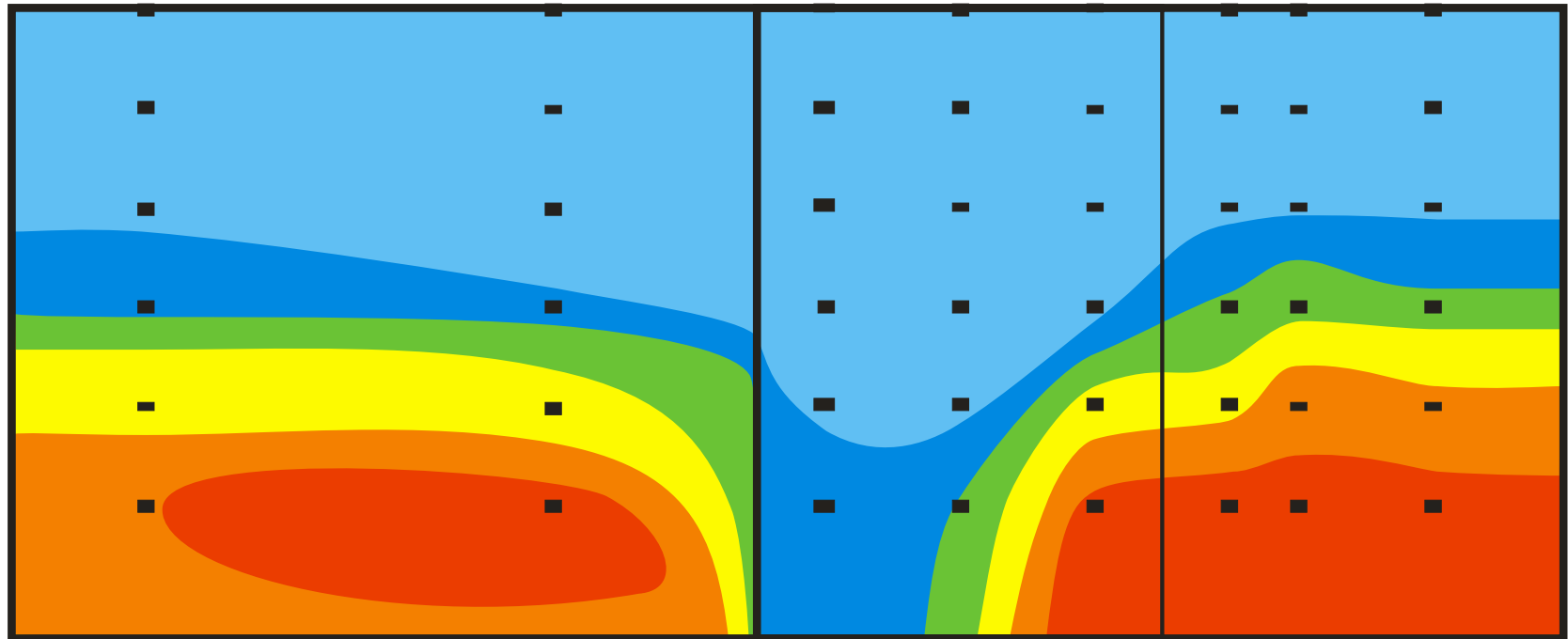
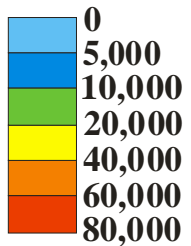
Alkalinity
(mg/L
CaCO₃)



Sulfate profile – 18 months

ground water flow →

SO₄ (mg/L)

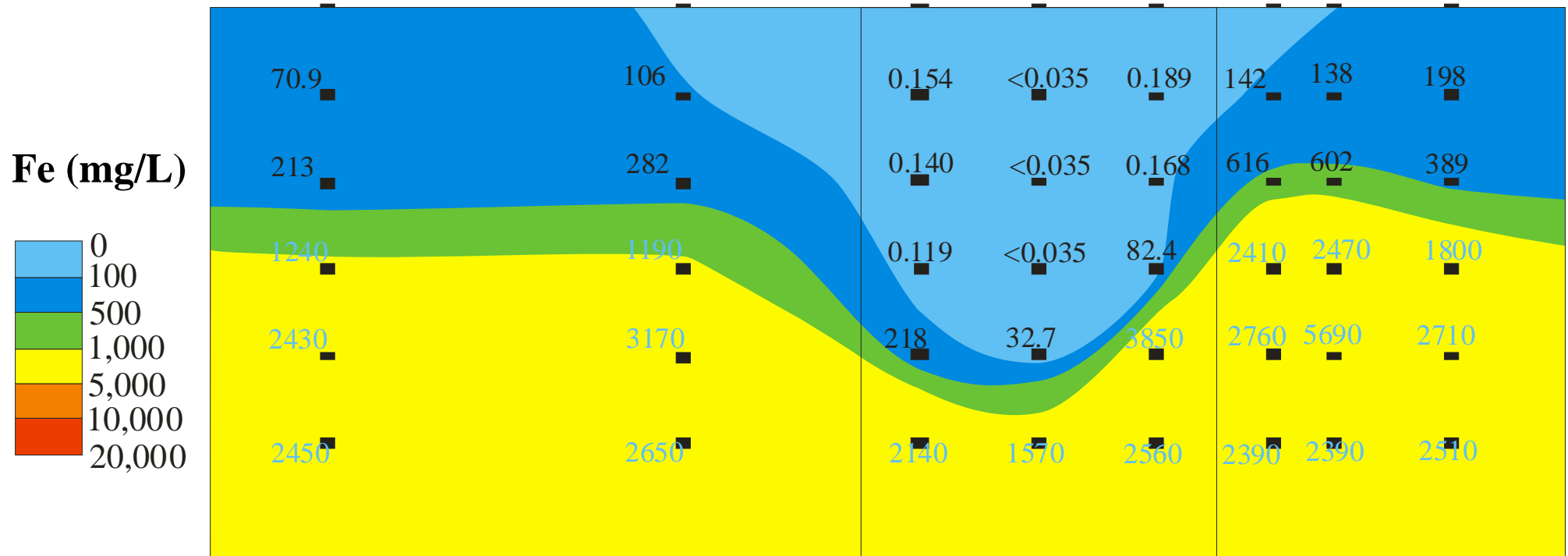


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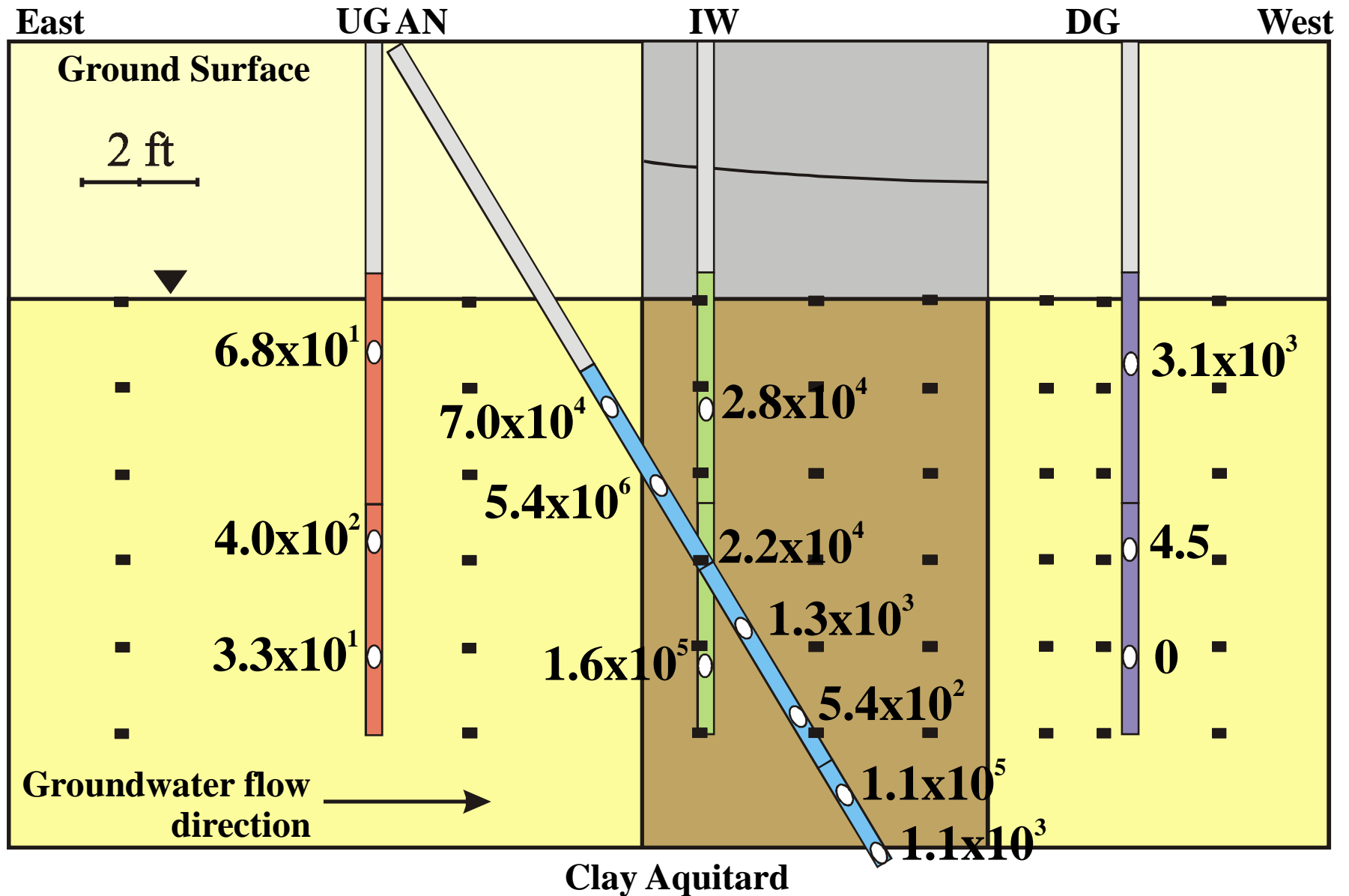
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Iron profile – 18 months

ground water flow 



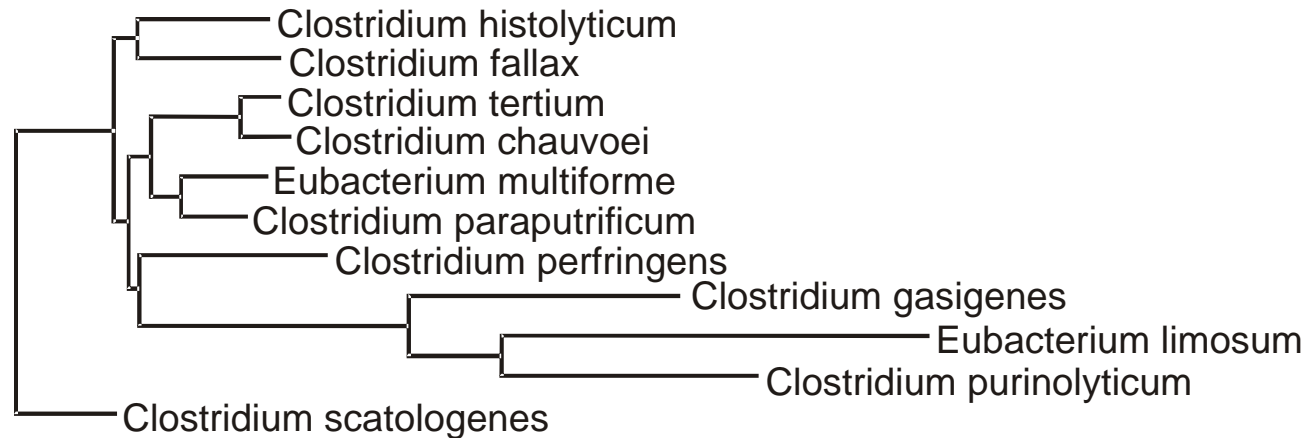
Sulfate reducers (MPN)



Phylogenetic Tree Display

Neighbor Joining Tree

N Join: 8.976 %



Alignment:

- 13.55 % 502 Clostridium tertium
- 14.34 % 502 Clostridium chauvoei
- 14.74 % 502 Eubacterium multiforme
- 14.75 % 517 Eubacterium limosum
- 14.84 % 516 Clostridium purinolyticum
- 15.00 % 500 Clostridium parapatrificum
- 15.54 % 502 Clostridium histolyticum
- 15.54 % 502 Clostridium fallax
- 15.74 % 502 Clostridium perfringens
- 16.20 % 500 Clostridium scatologenes



Summary

- 18-month data indicates effective removal of Pb, Cd, and As to date
- pH of influent ground water is increased from < 4 to > 6
- Ground water is converted from acid producing potential to acid consuming potential
- Sulfate reducers are present in PRB in significant numbers
- ^{34}S enrichment being observed in PRB ground water



Issues

- Hydraulics through PRB not yet fully resolved
- Ground water flow through much of site is very slow
- Upper part of aquifer appears to exhibit higher hydraulic conductivity than PRB



Future Work

- Continue semi-annual monitoring to assess performance and longevity of PRB
- Conduct tracer test to better evaluate hydraulics through PRB
- Evaluate solid phase chemistry within PRB
- Refine design of PRB for potential full-scale implementation

