

Remediation of Perchlorate, NDMA, and Chlorinated Solvents Using Nanoscale ZVI

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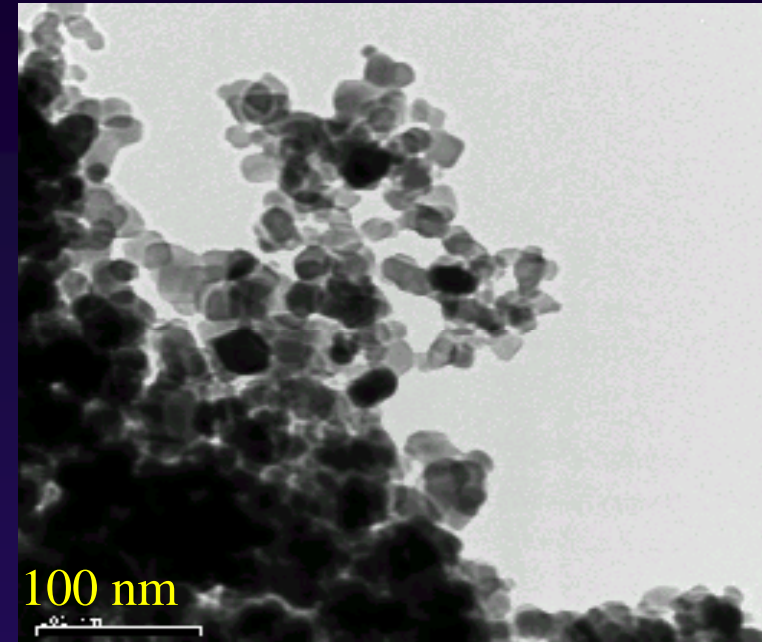
Presentation Outline

- ◆ Technology Overview
- ◆ Project Motivation
- ◆ Demonstration Project: California Site
 - ◆ Bench test approach
 - ◆ Bench test results
 - ◆ Pilot test plan
- ◆ R&D Needs



Nanoscale ZVI – Technology Overview

- ◆ Typical diameter < 100 nm
- ◆ Specific surface area: 33.5 m²/g (vs. < 1 m²/g for commercial Fe powder)
- ◆ Treatment rates 10 to 100 times faster than ZVI powder
- ◆ Addition of noble metal (< 1% wt.) dramatically increases treatment rates
- ◆ Ease of deployment:
 - ◆ Direct injection into wells
 - ◆ Ex situ slurry reactors (e.g., GAC, zeolite)
 - ◆ Fixed to membranes



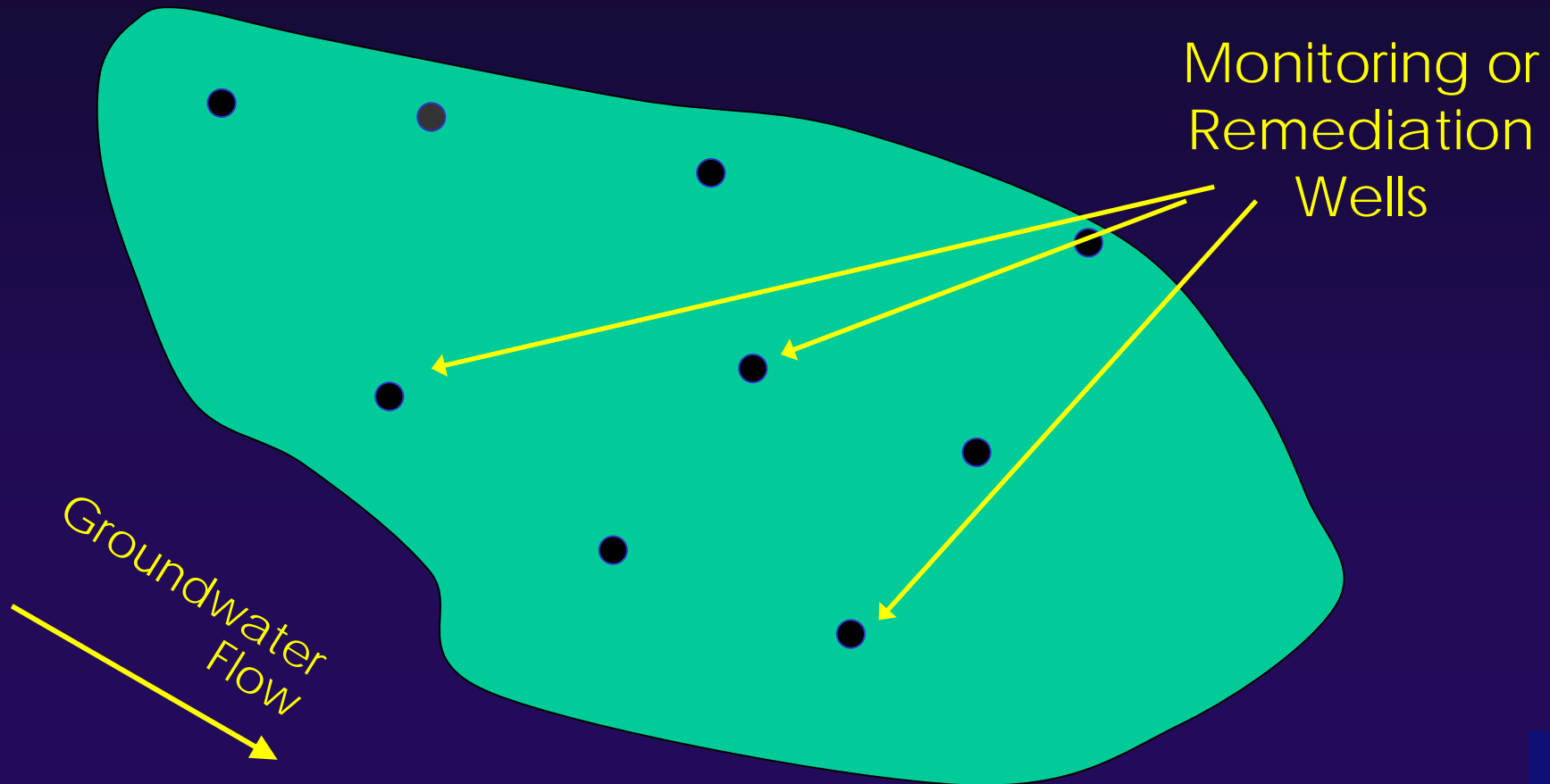
Nanoscale ZVI: Compounds Treated

- ◆ Chloroethenes – PCE, TCE, cDCE, VC
- ◆ Chloromethanes – CTET, Chloroform
- ◆ Chlorobenzenes
- ◆ Organochlorine pesticides (e.g., Lindane)
- ◆ PCBs
- ◆ N-nitrosodimethylamine (NDMA)
- ◆ Cr^{VI}, As^V, Pb^{II}
- ◆ Perchlorate, NO₃⁻
- ◆ *Everything else treated by granular ZVI, except > 10x faster*



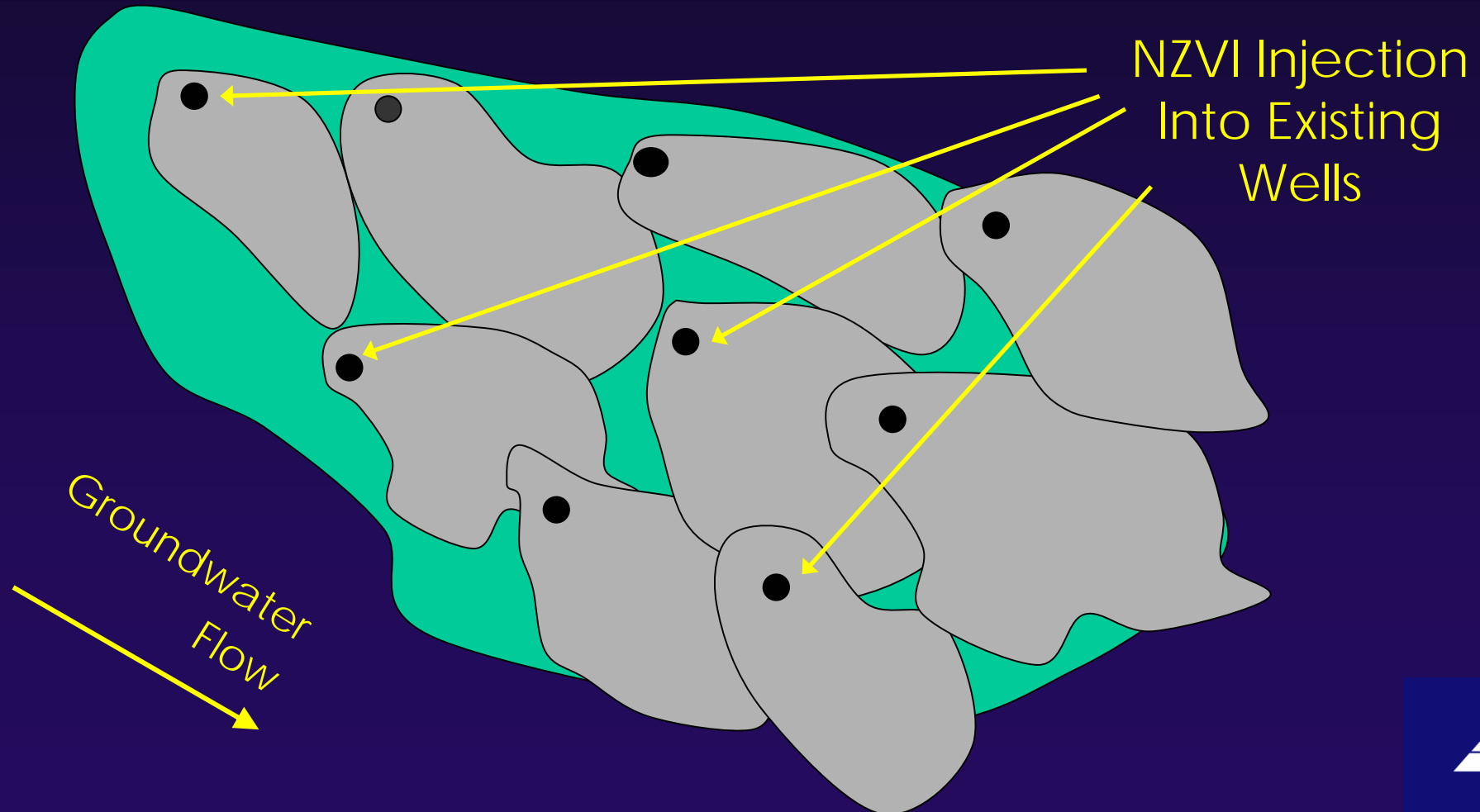
Nanoscale ZVI – In Situ Approaches

Plume-Wide Remediation



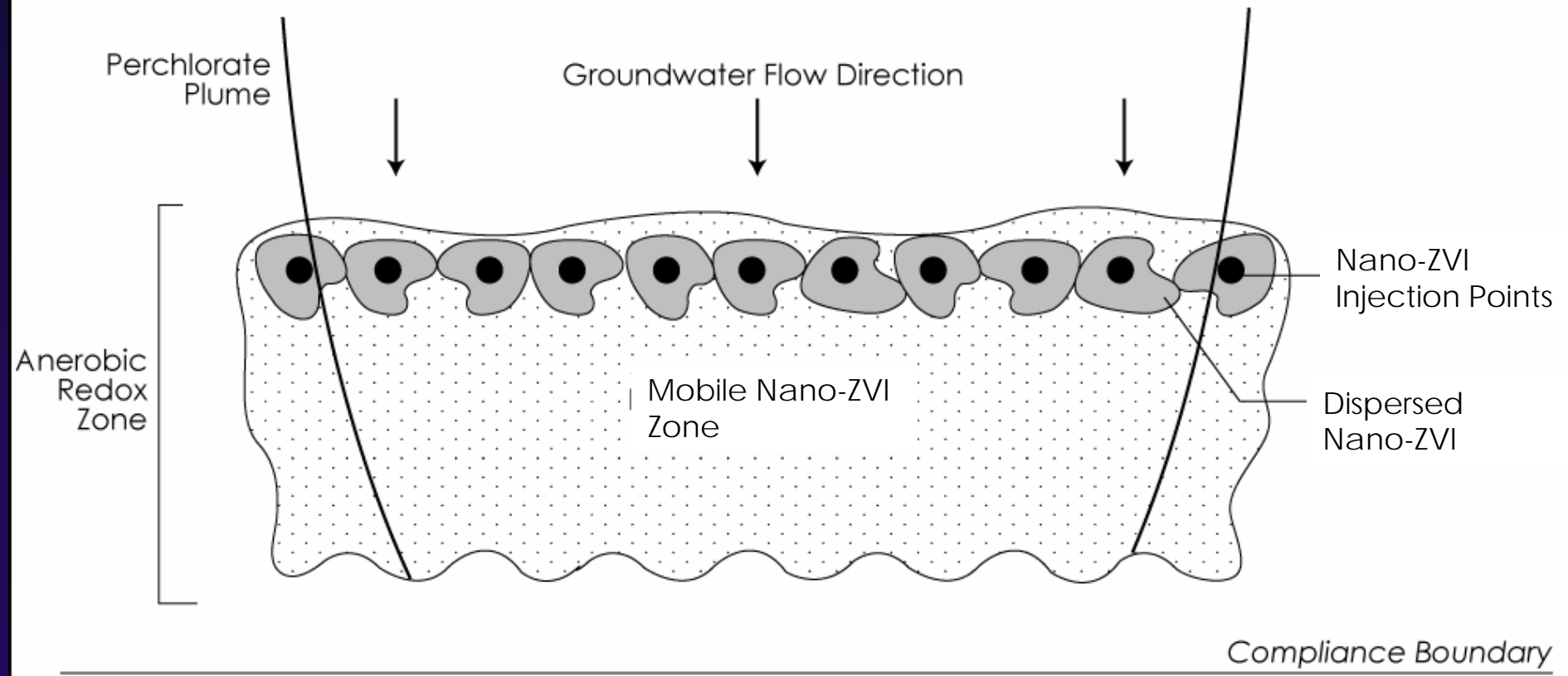
Nanoscale ZVI – In Situ Approaches

Plume-Wide Remediation



Nanoscale ZVI – In Situ Approaches

Plume Containment via PRB



Project Motivation

- ◆ Perchlorate (ClO_4) is a national threat to drinking water
 - Produced, handled, or used in at least 44 states
 - Present in groundwater supplies in ≥ 18 states
 - Detected in water supplies of > 15 M people in CA, AZ, NV
- ◆ Incidence of ClO_4 detection continues to increase
- ◆ ClO_4 is highly soluble and generally un-reactive in groundwater
- ◆ Plumes are frequently large and deep
- ◆ Conventional ex situ treatment technologies (e.g., ion exchange) carry high capital cost



Project Motivation

- ◆ ClO_4 plumes often contain *N*-nitrosodimethylamine (NDMA), and chlorinated solvents (especially at rocket test sites)
- ◆ Few (if any) conventional technologies can treat all these compounds at once
- ◆ Nanoscale ZVI has potential to treat these compounds simultaneously, in situ
- ◆ However, subsurface transport and longevity of NZVI poorly understood



Project Motivation

- ◆ Moore et al. (2003) demonstrated abiotic dechlorination of ClO_4 in presence of granular and powder ZVI, and
- ◆ Concluded that rates were too slow for application for site remediation...But,
- ◆ Did not examine nanoscale ZVI

Rate and Extent of Aqueous Perchlorate Removal by Iron Surfaces

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Environ. Sci. Technol. 2003, 37, 3189–3198



Project Motivation

- ◆ Gui et al. (2000) demonstrated effective abiotic dechlorination of NDMA in presence of granular ZVI, and Nickel-ZVI;
- ◆ Transformation rates have not been reported for Nano-ZVI

Reduction of *N*-Nitrosodimethylamine with Granular Iron and Nickel-Enhanced Iron. 1. Pathways and Kinetics

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Environ. Sci. Technol. 2000, 34, 3489–3494



**Technology Demonstration:
Treatment of ClO_4 , NDMA,
Chloroethenes, and Nitrate
at Aerospace Site**

Sacramento, California



Demonstration Objectives

Phase I: Bench Test (On-going)

- ◆ Confirm treatability
- ◆ Determine site-specific Nano-ZVI loading requirements
- ◆ Estimate site-specific transformation rates
- ◆ Determine treatment capacity (reactive longevity) under site-specific conditions
- ◆ Measure retardation and transport of Nano-ZVI in aquifer columns

Phase II: Pilot Test (Planned)

- ◆ Nano-ZVI delivery and performance in active recirculation cell
- ◆ Nano-ZVI delivery and performance in passive PRB



Example Batch Reactors



Optimal Loading Tests (ongoing)

Objective: estimate optimal range for NZVI loading concentration

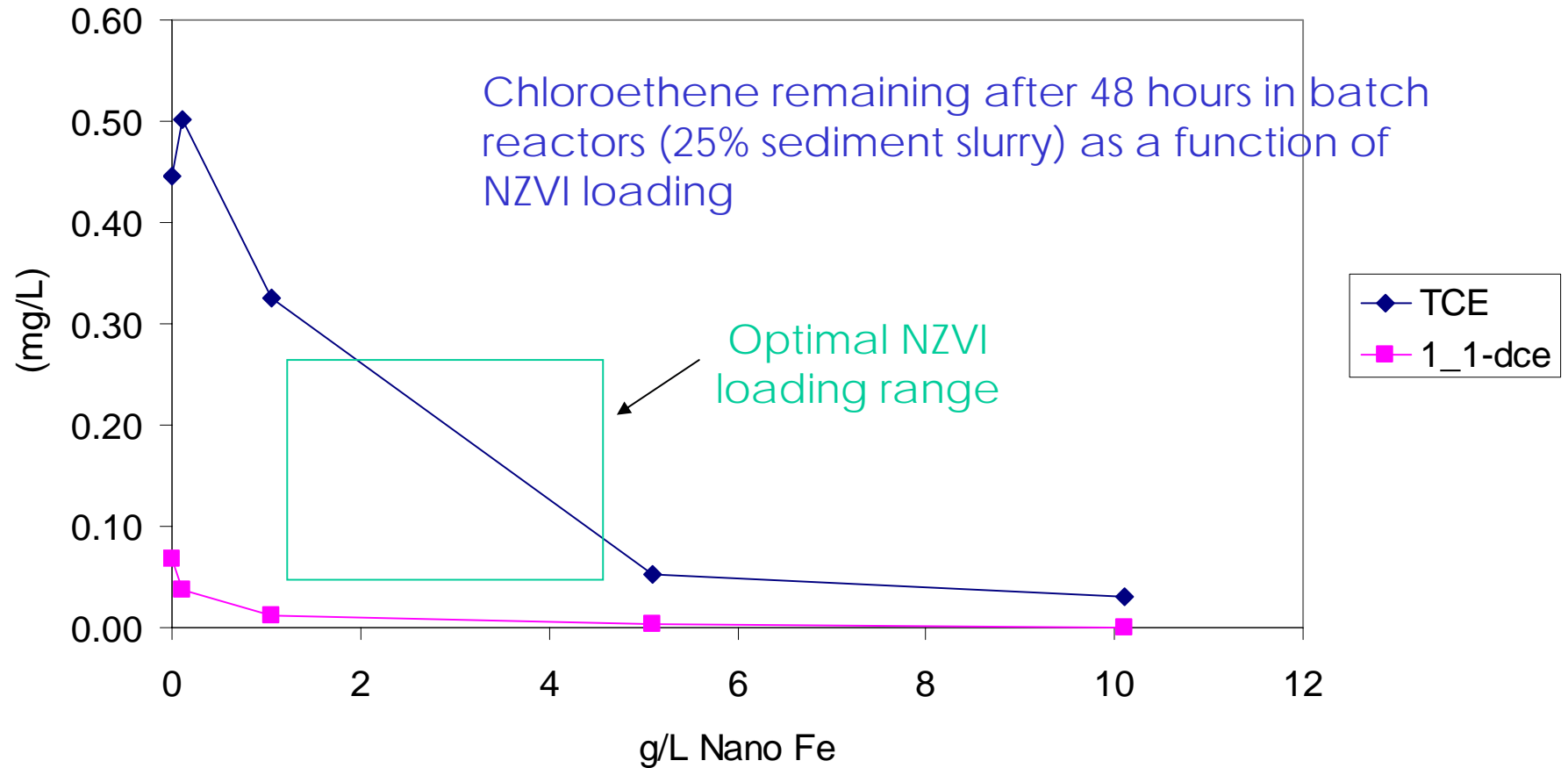
Set-up

- ◆ Separate batch reactors treated with either 0, 0.5, 1, 5, or 10 g/L Nano-ZVI
- ◆ 25% aquifer sediment slurry in site groundwater
- ◆ React for 48 hours
- ◆ Measure contaminant remaining
- ◆ Only tested chloroethenes and NDMA so far...



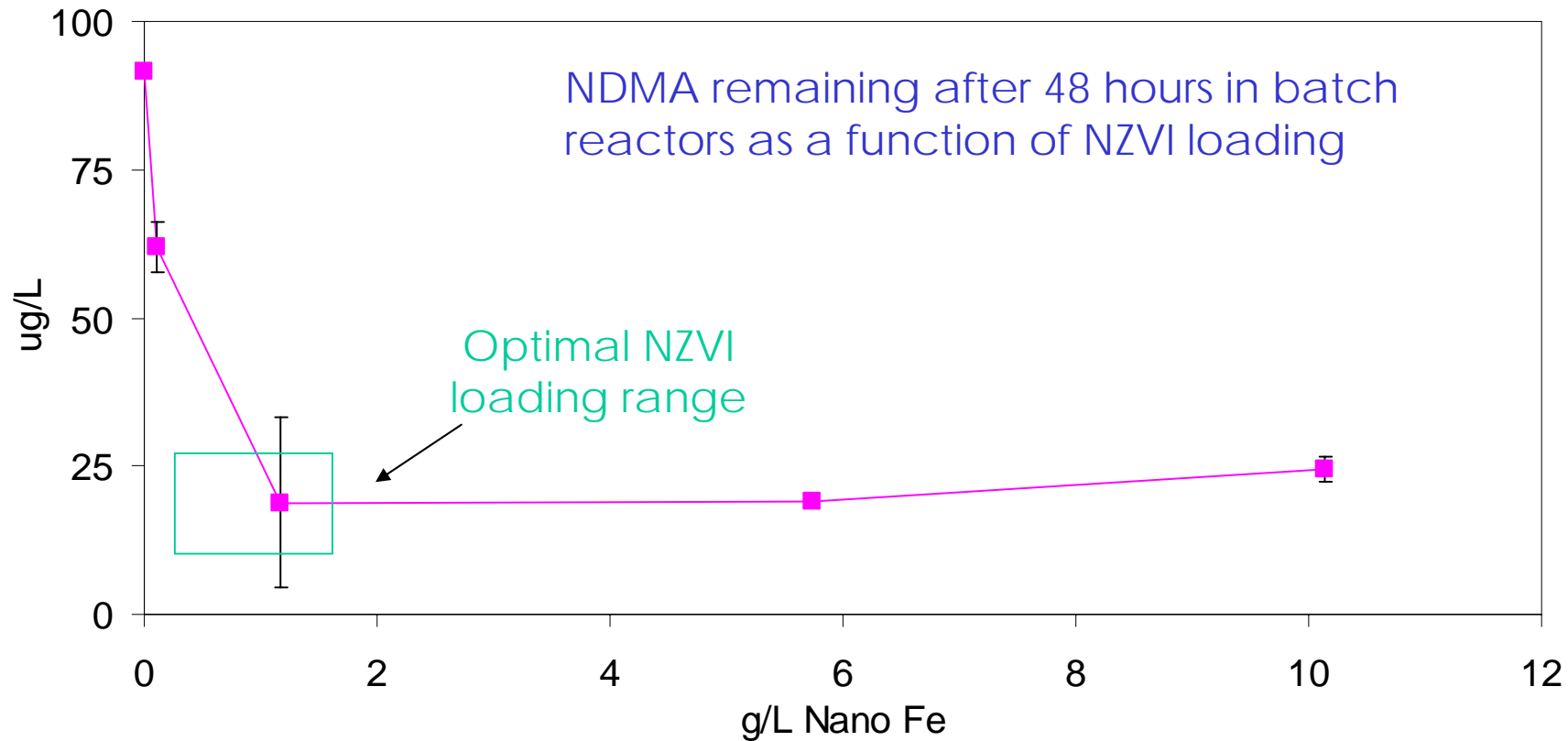
Preliminary Results

Determination of Optimal Loading for Chloroethene Treatment



Preliminary Results

Determination of Optimal NZVI Loading For NDMA Treatment



Optimal Loading Tests – Summary of Preliminary Results

- ◆ Nano-ZVI effective treatment agent for TCE, NDMA
- ◆ TCE and perchlorate are driver compounds (loading requirement for TCE \gg NDMA)
- ◆ Optimal Nano-ZVI loading requirement: 2 to 5 g/L



Transformation Rate Tests

Objectives

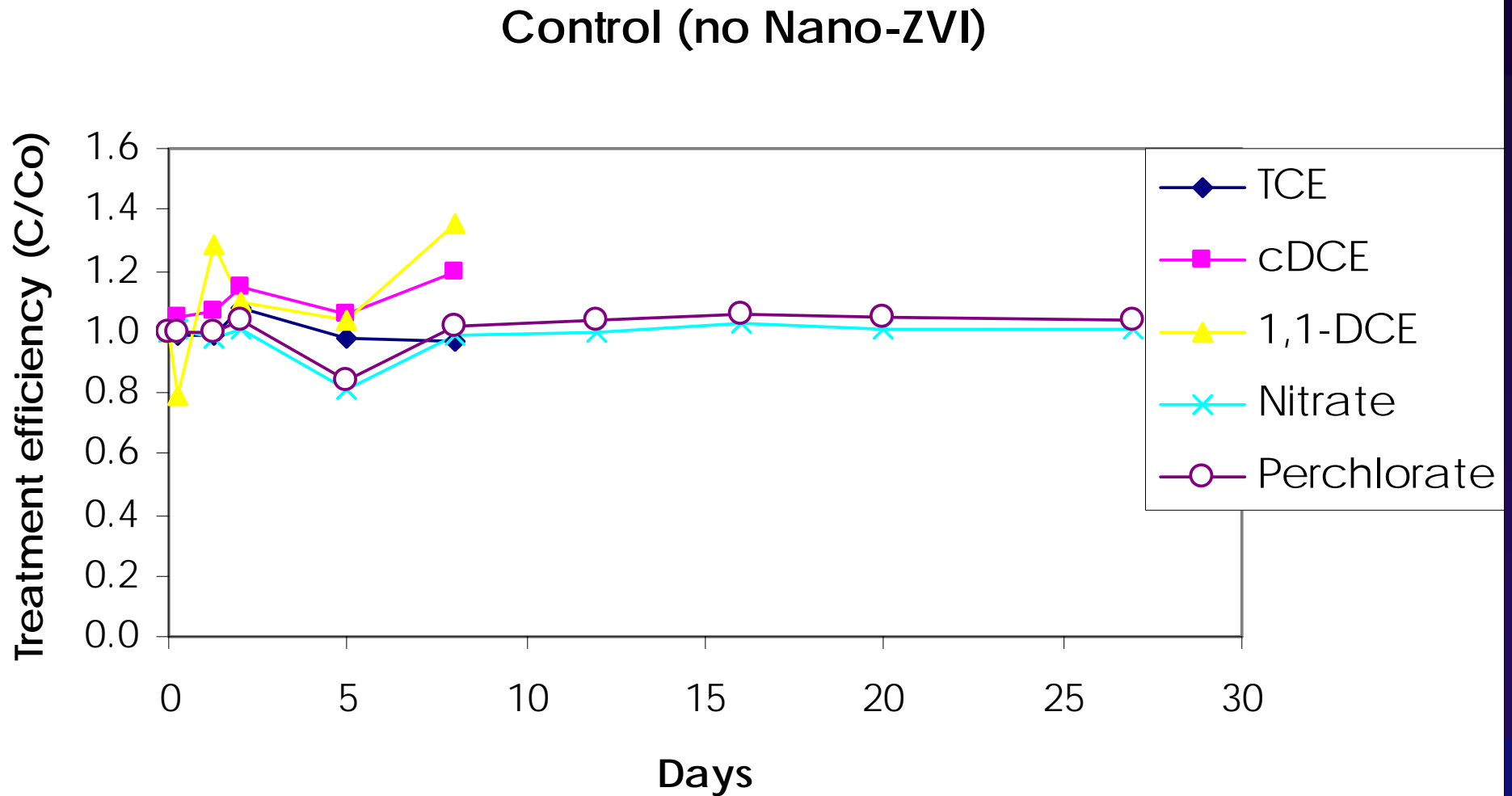
- (1) estimate ClO_4^- , chloroethene, and nitrate transformation rates in the presence of 10 g/L Nano-ZVI
- (2) compare rates for commercial and laboratory synthesized NZVI

Set-Up

- ◆ Duplicate reactors, sampled at various intervals for 27 days
- ◆ 25% aquifer sediment slurry in site groundwater
- ◆ Included control reactors (sediment + water, w/o Nano-ZVI)
- ◆ Initial concentrations: $\text{ClO}_4^- = 14$ ppm; TCE = 2 ppm; nitrate = 20 ppm

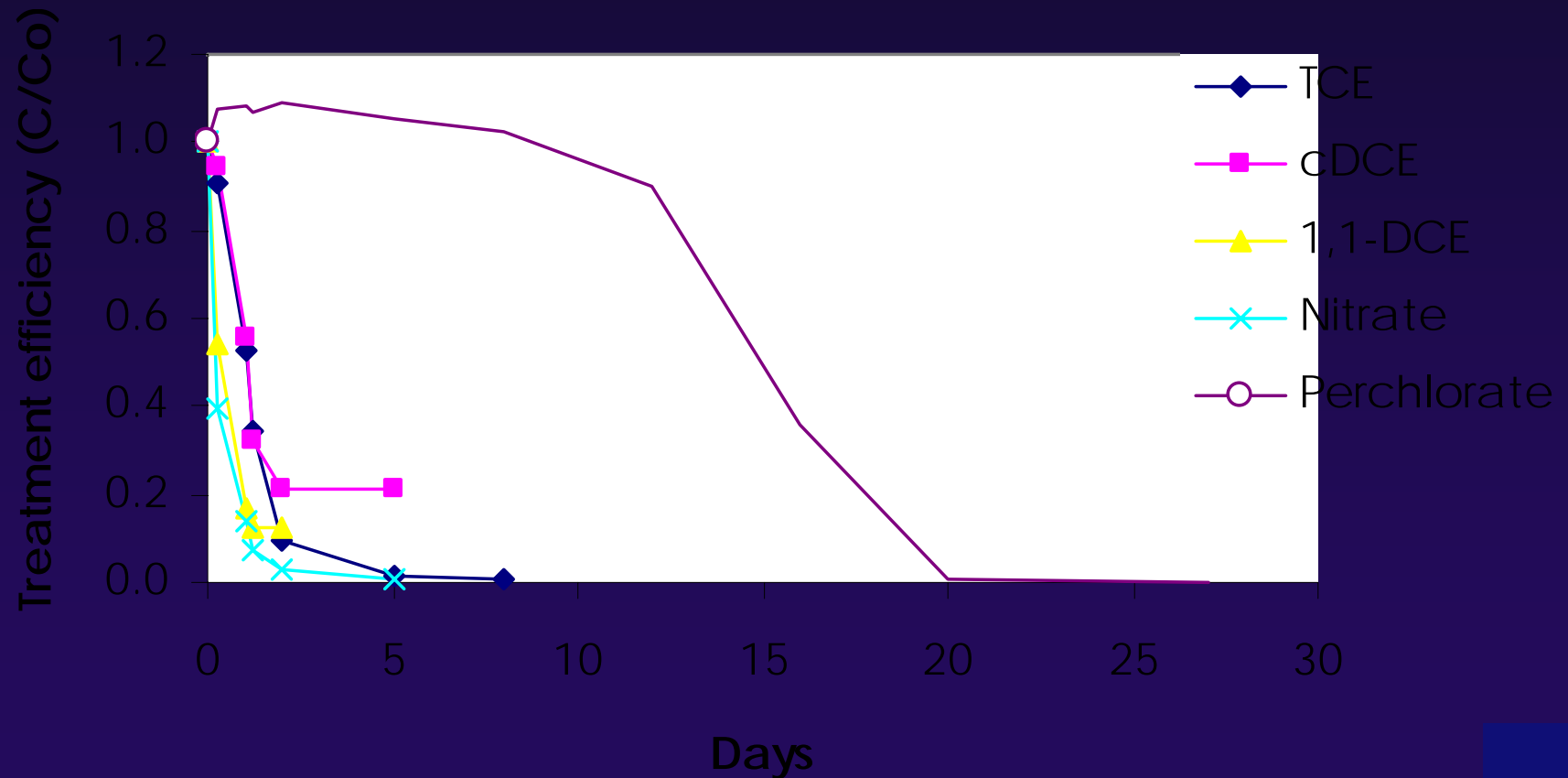


Transformation Rate Tests – Controls

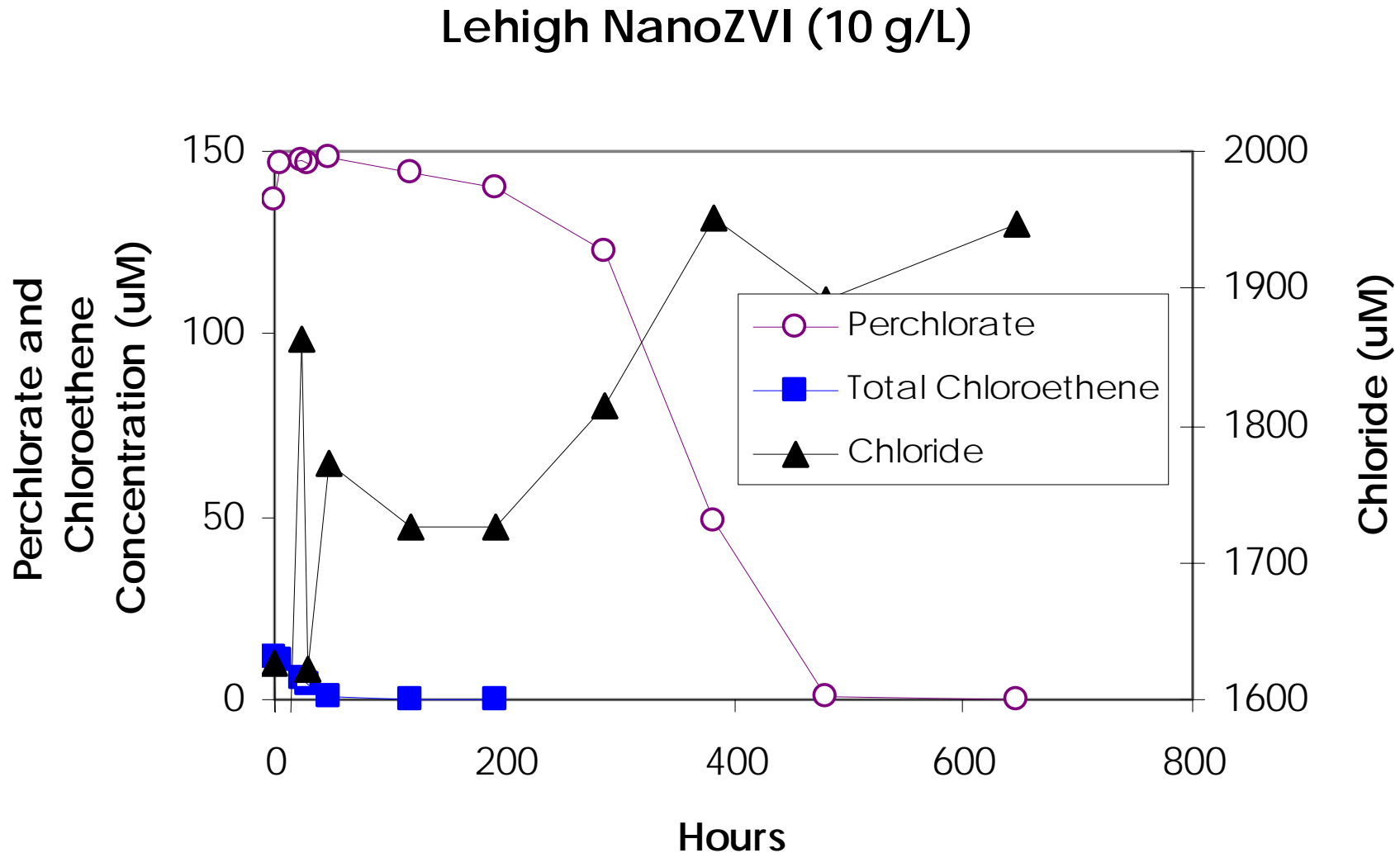


Transformation Rate Tests Lehigh Nano-ZVI

Lehigh NanoZVI (10 g/L)



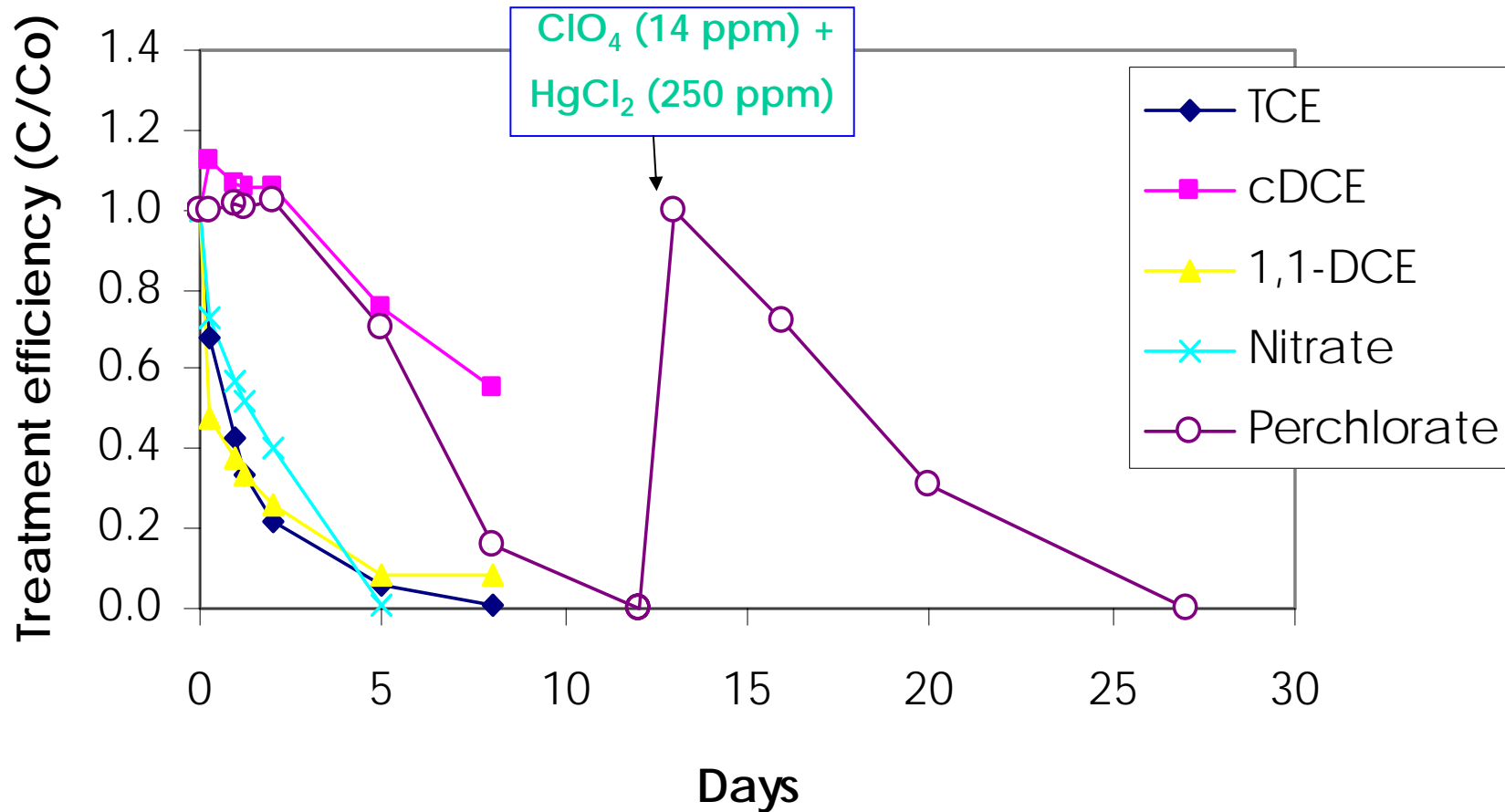
Transformation Rate Tests Lehigh Nano-ZVI



Transformation Rate Tests

Toda Nano-ZVI

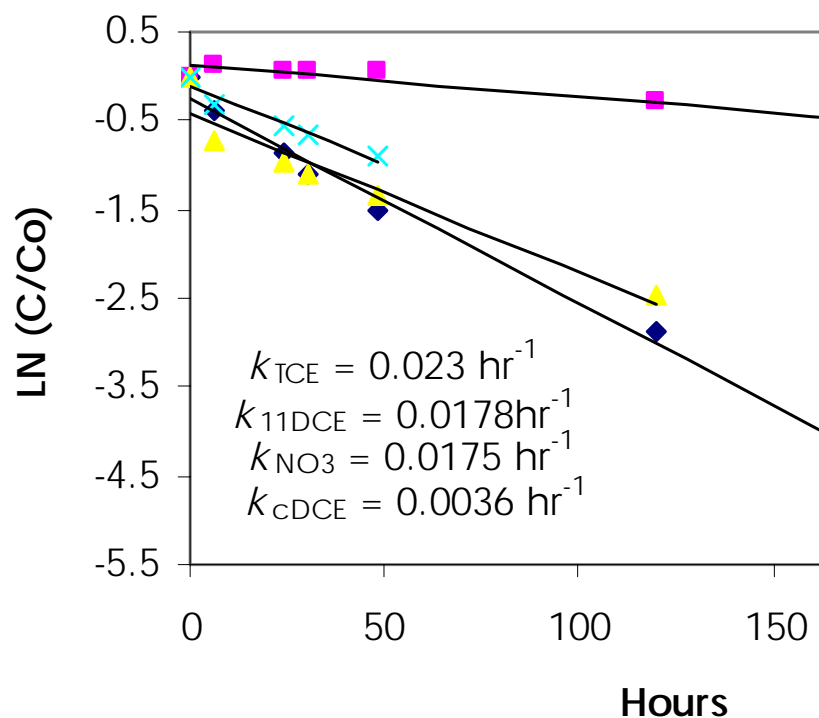
Toda NanoZVI (10 g/L)



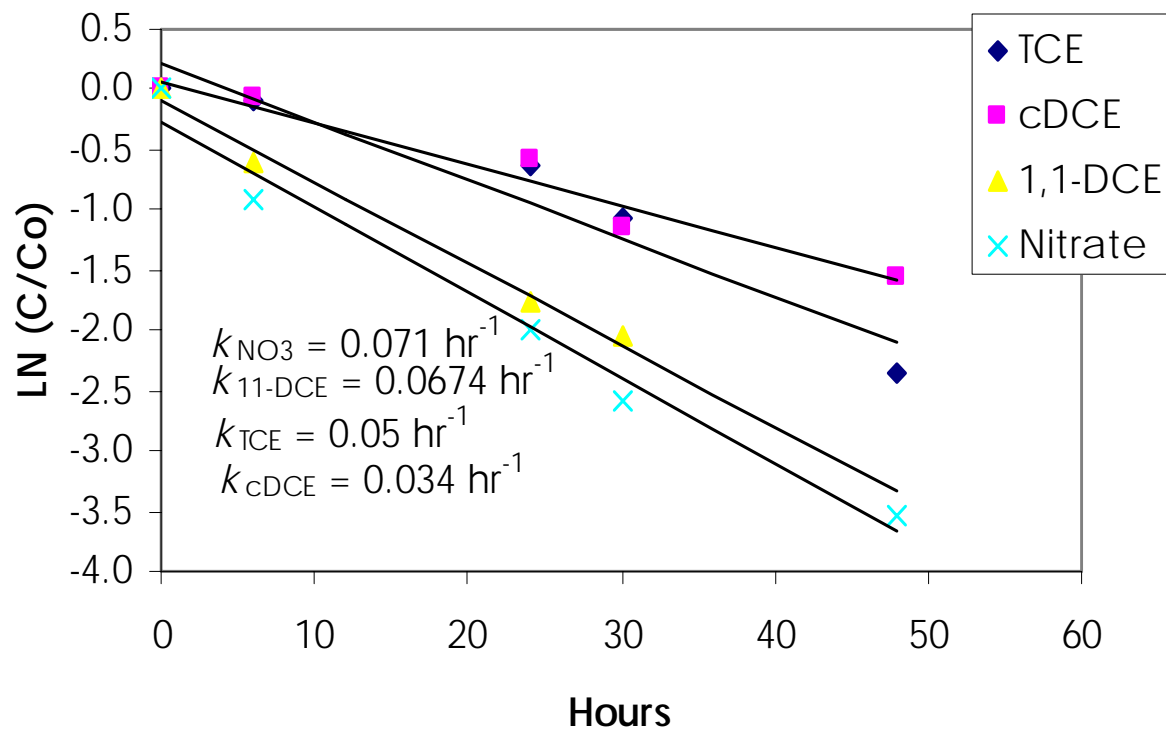
Transformation Rate Tests

Comparison of Nano-ZVI Performance for Chloroethenes

Toda NanoZVI (10 g/L)



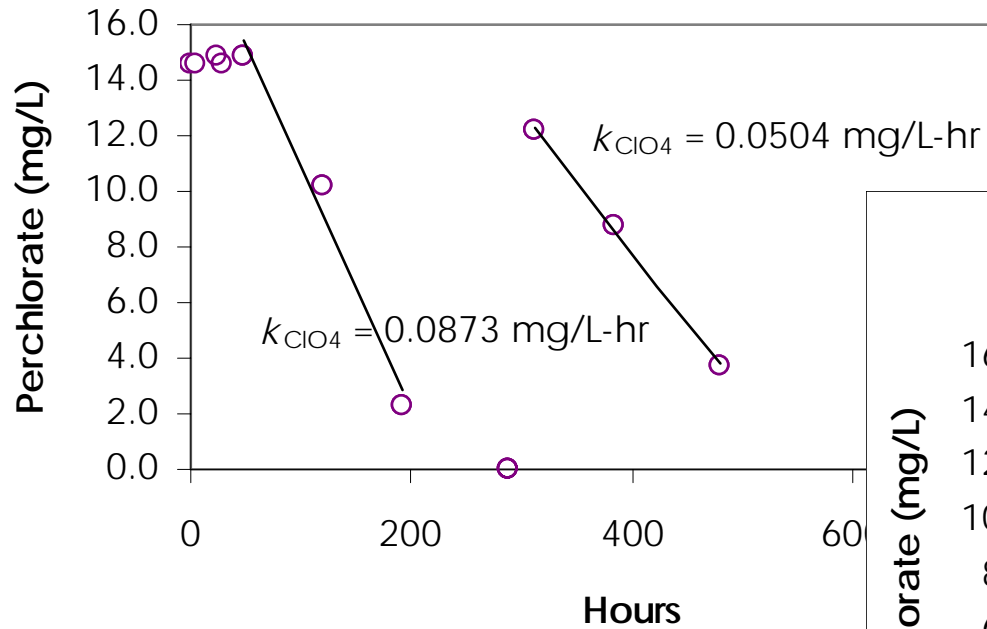
Lehigh NanoZVI (10 g/L)



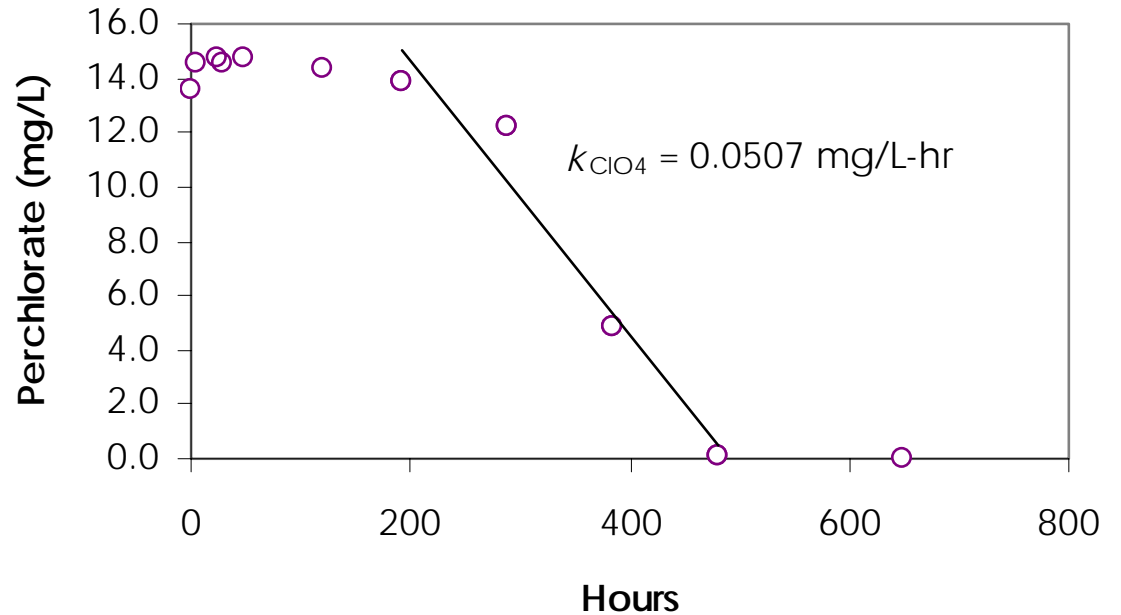
Transformation Rate Tests

Comparison of Nano-ZVI Performance for Perchlorate

Toda NanoZVI (10 g/L)



Lehigh NanoZVI (10 g/L)



Transformation Rate Test – Summary of Observed Transformation Rates

Constituent	Toda Nano-ZVI				Lehigh Nano-ZVI			
	Zero Order (mg/L-hr)	First Order (hr ⁻¹)	Half-Life (d)	Surface-Area Normalized Rate Coefficient (L/hr/m ²)	Zero Order (mg/L-hr)	First Order (hr ⁻¹)	Half-Life (d)	Surface-Area Normalized Rate Coefficient (L/hr/m ²)
Perchlorate	0.0504				0.0507			
TCE		0.023	1.3	6.57E-05		0.050	0.6	1.49E-04
1,1-DCE		0.018	1.6	5.09E-05		0.067	0.4	2.01E-04
c-DCE		0.004	8.0	1.03E-05		0.034	0.8	1.01E-04
Nitrate		0.018	1.7	5.00E-05		0.071	0.4	2.12E-04



Transformation Rate Test – Summary of Preliminary Findings

- ◆ Nano-ZVI treats chloroethenes, ClO_4 , NDMA, and nitrate in Site groundwater; ClO_4 is rate-limiting step.
- ◆ Lehigh Nano-ZVI degraded chloroethenes 2 to 3 times faster than Toda Nano-ZVI.
 - Why? Lehigh material has fewer impurities (i.e., oxides)
- ◆ Degradation of ClO_4 was preceded by lag period for both for Toda NZVI and Lehigh NZVI
 - Why? Possible that anions competitively inhibited ClO_4 sorption to Nano-ZVI (this is a critical first step)



Transformation Rate Test – Summary of Preliminary Findings

- ◆ Lag period to onset of ClO_4 degradation was shorter for Toda NZVI (2 to 5 days) than for Lehigh NZVI (5 to 8 days)
 - Why? Mixed oxide/ZVI content of Toda-NZVI is more favorable for dechlorination of perchlorate (Moore et al. 2003)

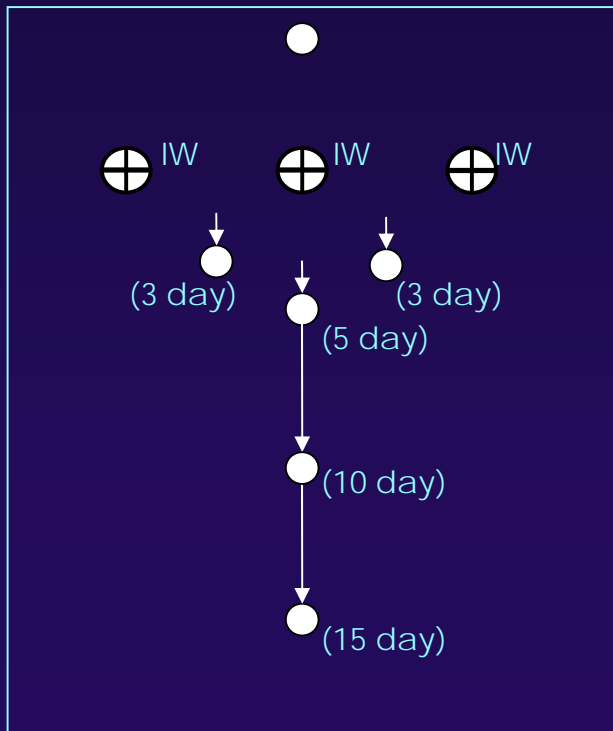


Aerospace Site – Pilot Test Plans

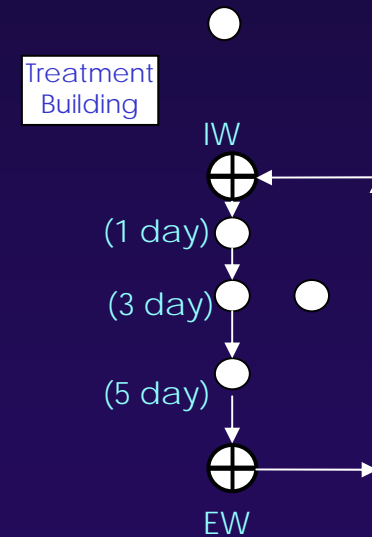
Direction of
Groundwater Flow



Passive Treatment Demo Layout

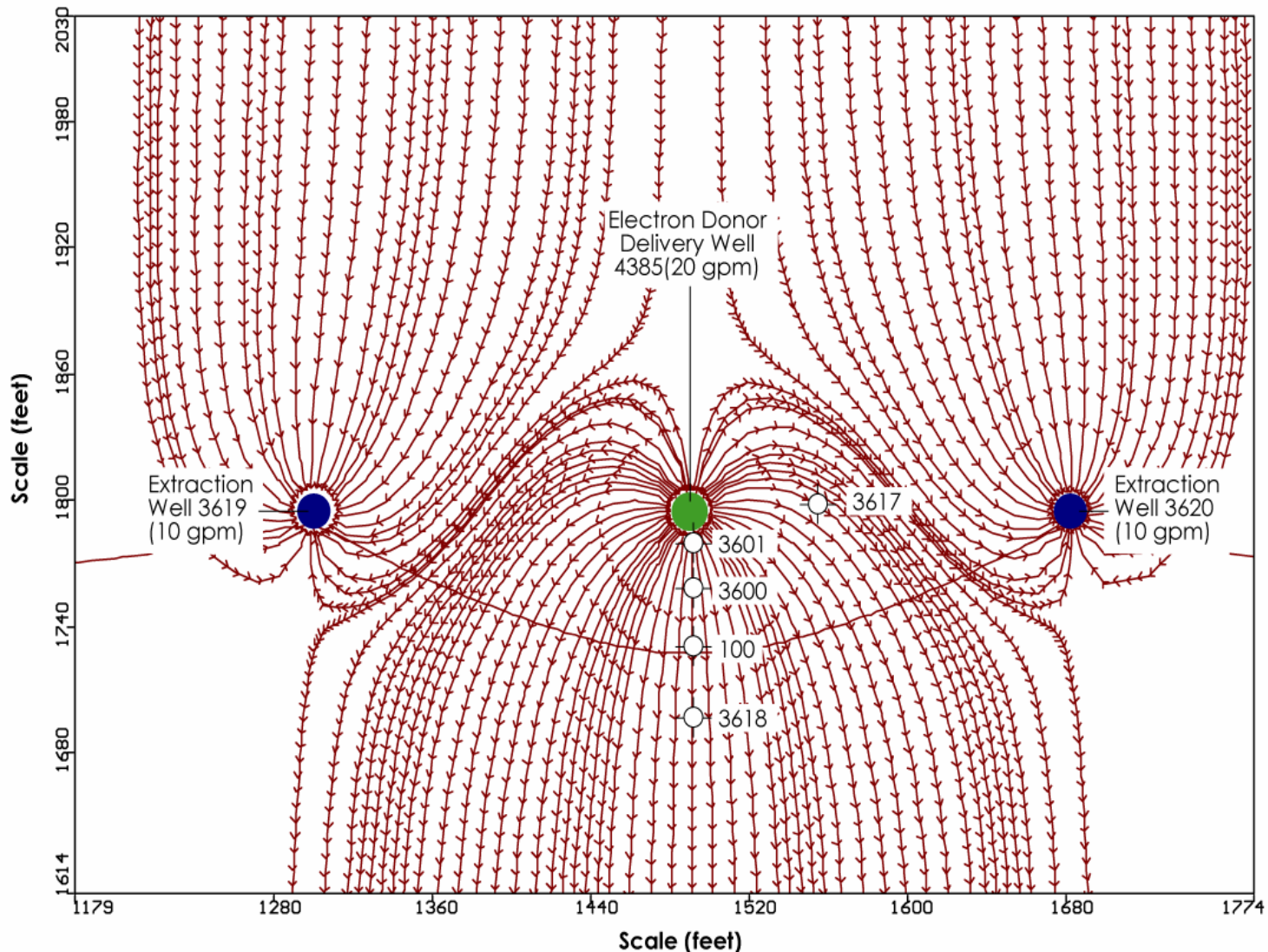


Active Treatment Demo Layout



Aerospace Site

Potential Approach for NZVI delivery



Research and Development Needs

- ◆ Cost effective approaches for particle synthesis/manufacture
- ◆ Field studies evaluating Nano-ZVI particle transport and reactive longevity
- ◆ Laboratory evaluation of relationship between particle shape and transport characteristics
- ◆ Evaluation of optimal modes for particle delivery
- ◆ Rigorous performance and cost comparison with in situ bioremediation (for perchlorate and chloroethenes)

